

**NANYANG  
TECHNOLOGICAL  
UNIVERSITY**

17 April, 2008

# **Catalytic Reactions in Organic Synthesis**

## **– Rhodium Complexes and Redox Catalysts –**

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School of Physical and Mathematical Sciences

# Rhodium Complexes

Electron Deficient-Lower Valent Rh Complexes

Bimetallic Rh Complexes

# Redox Catalysts

Metal Catalyzed One-Electron Transfer

Organic Electron Transfer Reagents

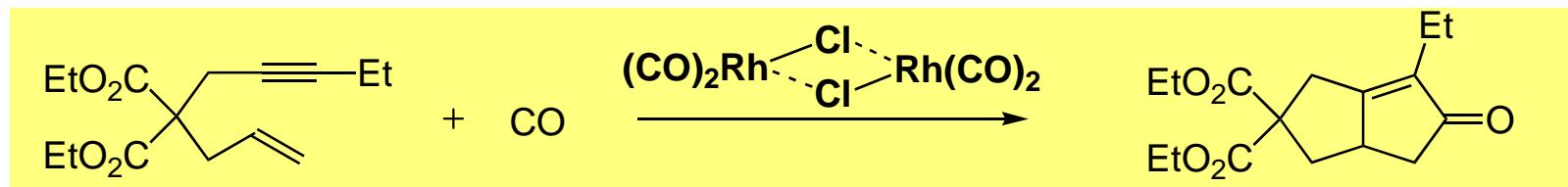
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Brønsted & Lewis Acids and Bases

Metal Catalysts for Coupling, Addition, etc.

Organo-Catalysts

Biocatalysts

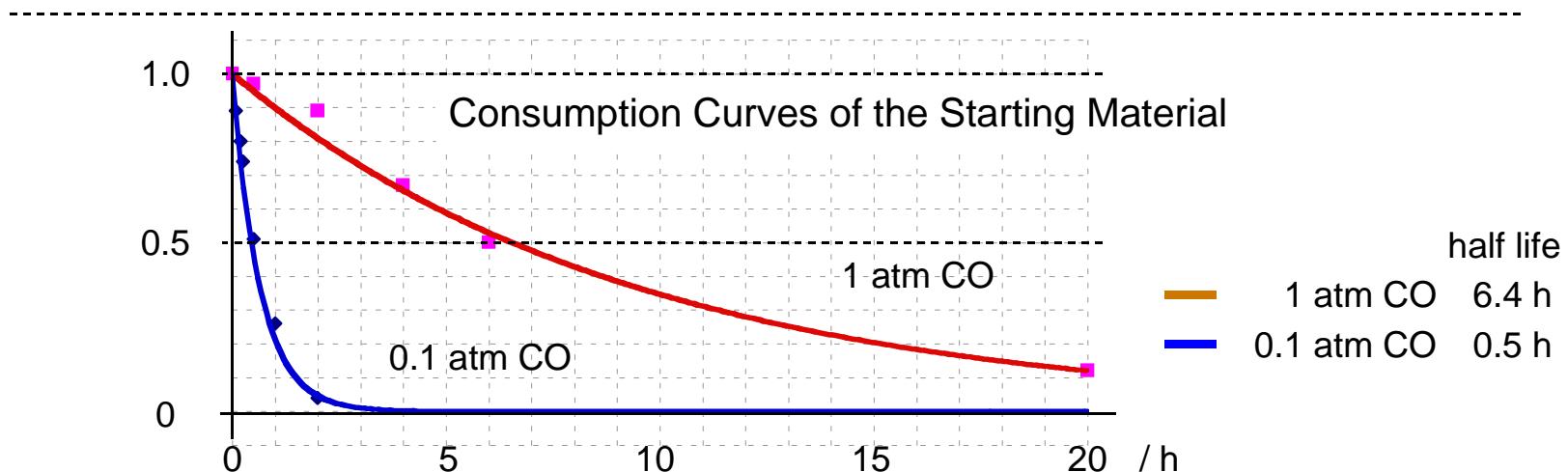


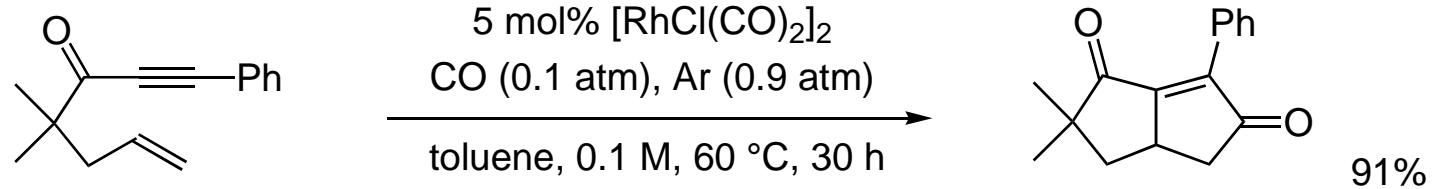
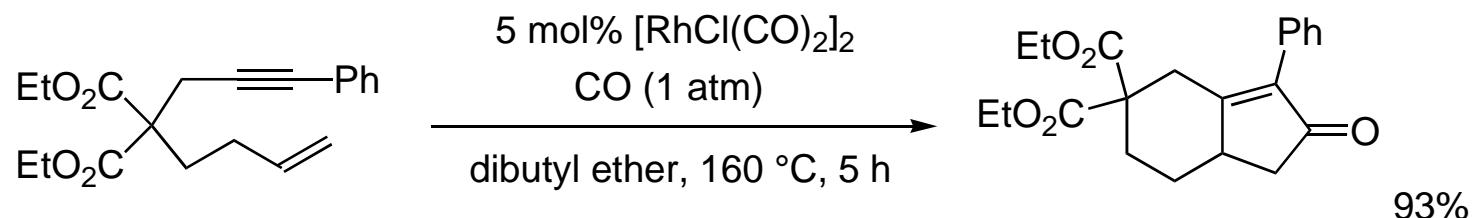
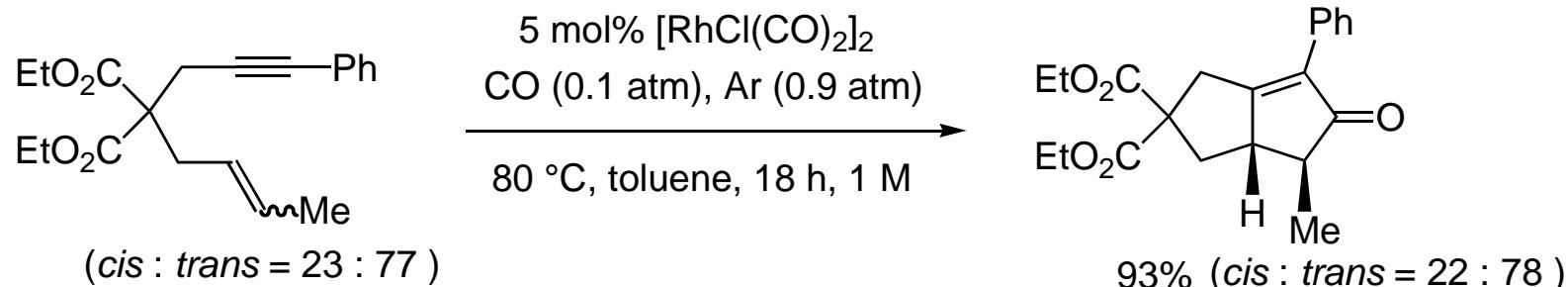
### Catalytic Intramolecular Pauson–Khand Reactions (~1997)

			Catalyst	Solvent	CO / atm	Temp. / °C
Co	Jeong and Chung	2 mol%	Co(cod)(indenyl)	DME	15	100
	Jeong and Chung	3 mol%	Co <sub>2</sub> (CO) <sub>8</sub> / P(OPh) <sub>3</sub>	DME	3	120
	Chung	5 mol%	Co(acac) <sub>2</sub> / NaBH <sub>4</sub>	CH <sub>2</sub> Cl <sub>2</sub>	30-40	100
	Jeong	2.5-5 mol%	Co <sub>2</sub> (CO) <sub>8</sub>	scCO <sub>2</sub>	110-120	90
	Chung	1 mol%	Co <sub>4</sub> (CO) <sub>12</sub>	CH <sub>2</sub> Cl <sub>2</sub>	10	150
	Livinghouse	5 mol%	Co <sub>2</sub> (CO) <sub>8</sub> / <i>hν</i>	DME	1	50-55
Ti	Buchwald	5-20 mol%	CpTi(CO) <sub>2</sub>	Toluene	1	90
	Chung	5-20 mol%	(chiral Cp)Ti(CO) <sub>2</sub>	Toluene	1	90
Ru	Murai	2 mol%	Ru <sub>3</sub> (CO) <sub>12</sub>	dioxane	10	160
	Mitsudo	2 mol%	Ru <sub>3</sub> (CO) <sub>12</sub>	DMAc	15	140



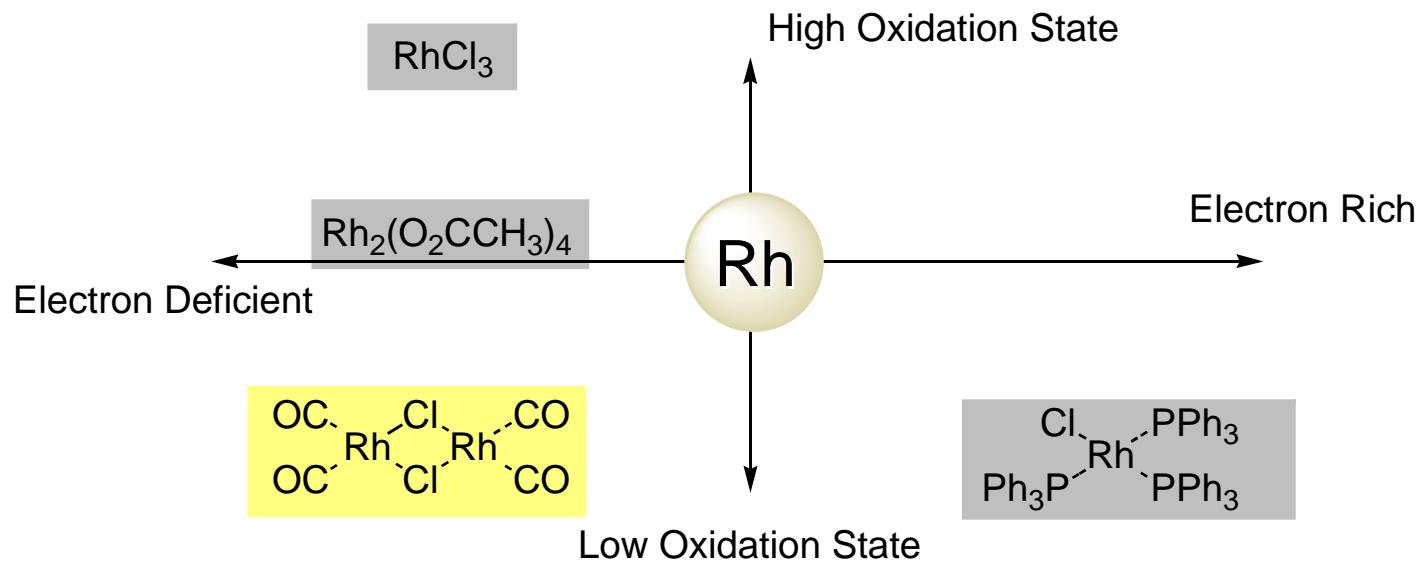
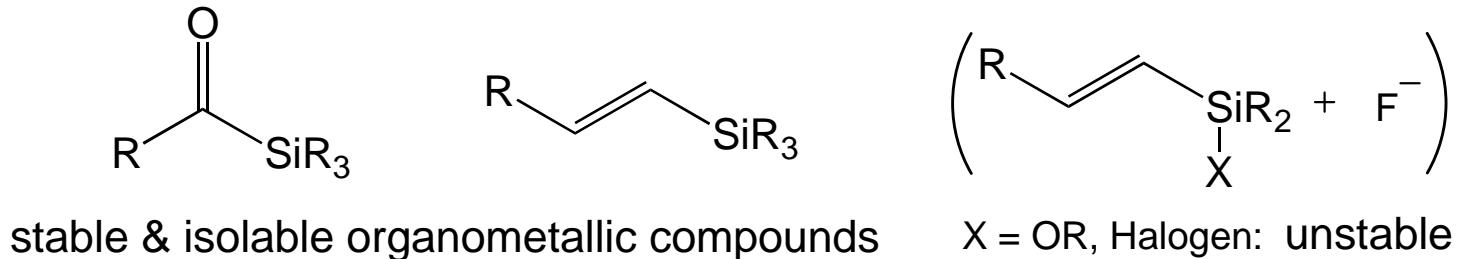
pressure / atm CO	Ar	time / h	yield / %	(recovery / %)
10	0	18	trace	(ca. 100)
3	0	36	70	(24)
1	0	18	91	—
0.2	0.8	5	90	—
0.1	0.9	2	92	—
0.05	0.95	3	85	—

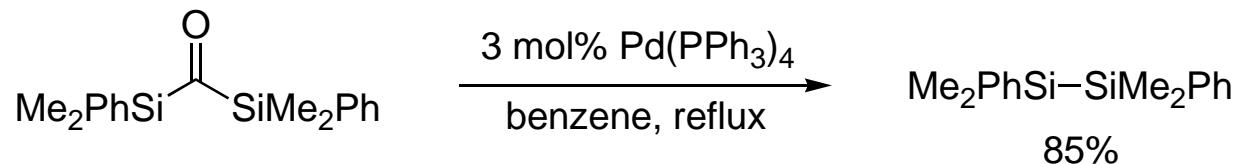
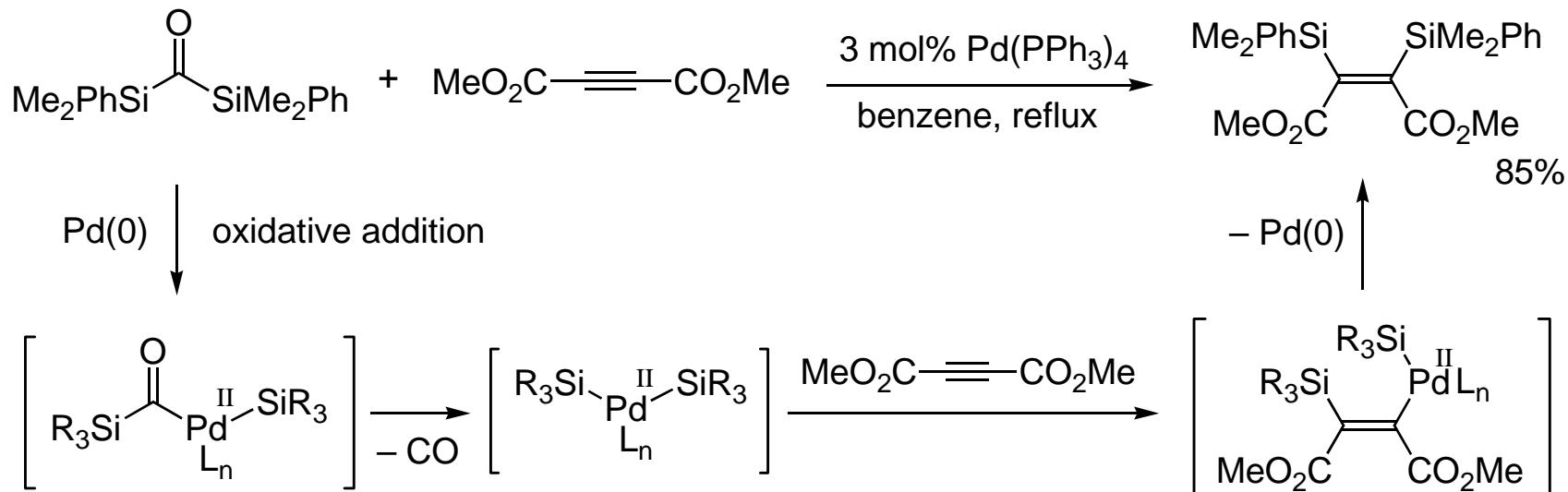




*J. Organomet. Chem.* **2001**, *624*, 73.

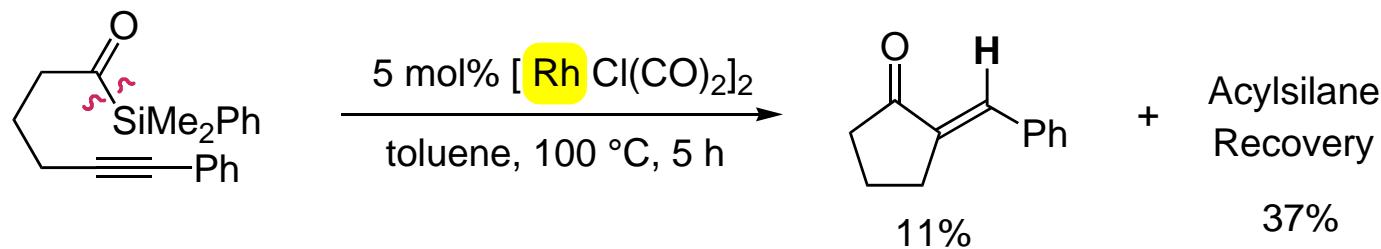
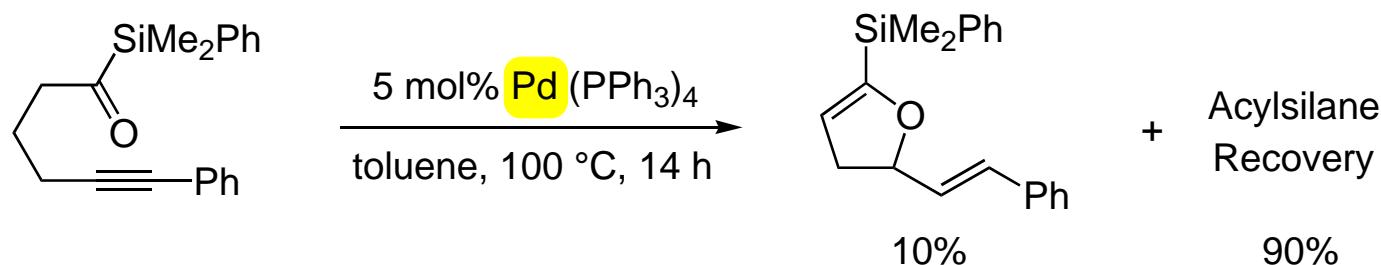
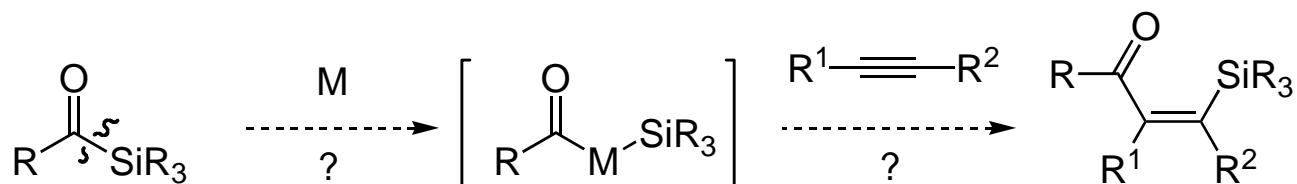
## Activation of Acyl and Vinyl Silanes with $[\text{RhCl}(\text{CO})_2]_2$

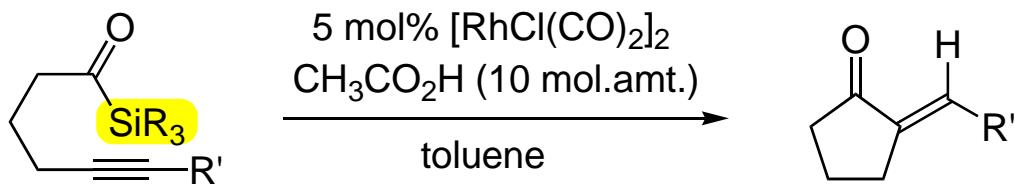




*Chem. Lett.*, 1996, 841.

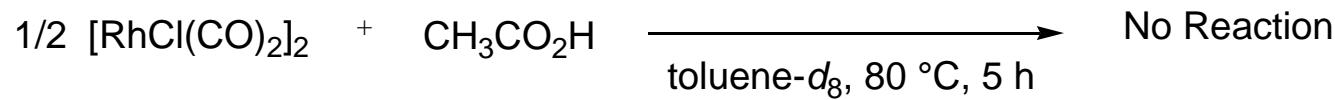
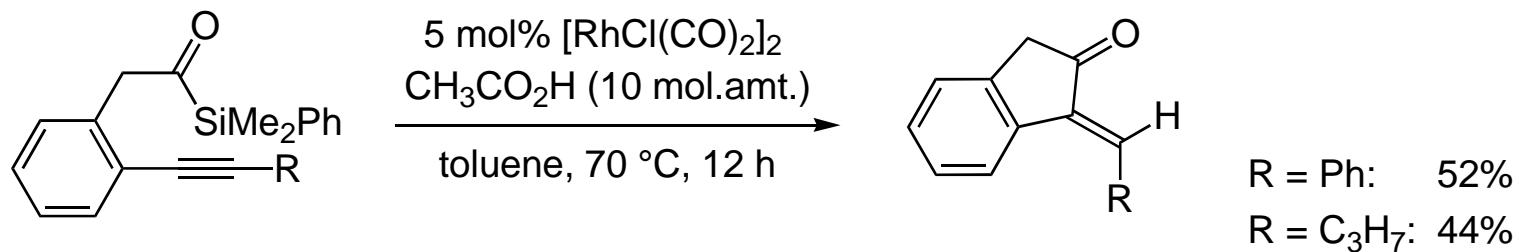
## Acylsilanes (Air-Stable Acyl Main Group Metal Compounds) (2001 –)

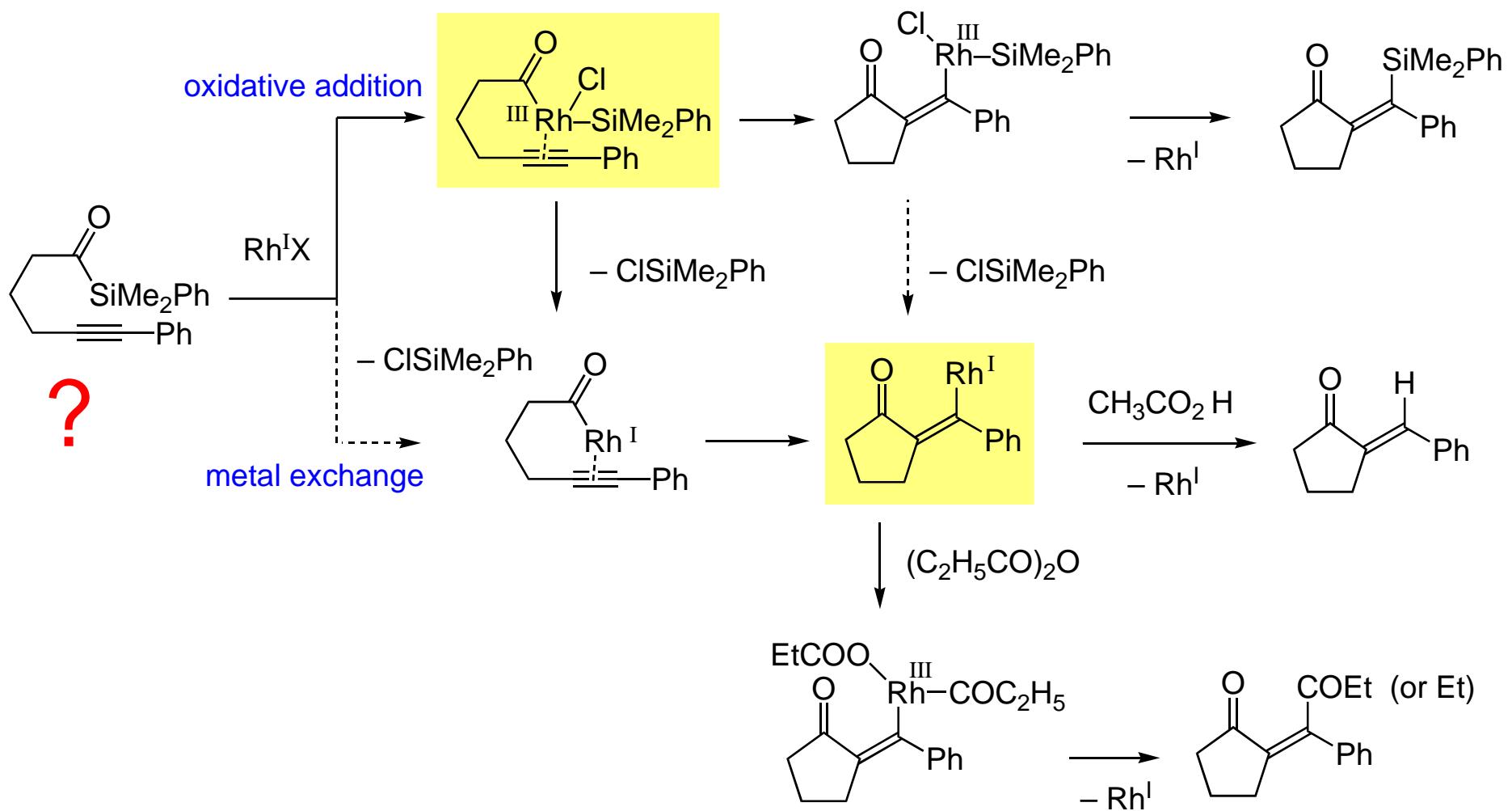




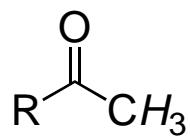
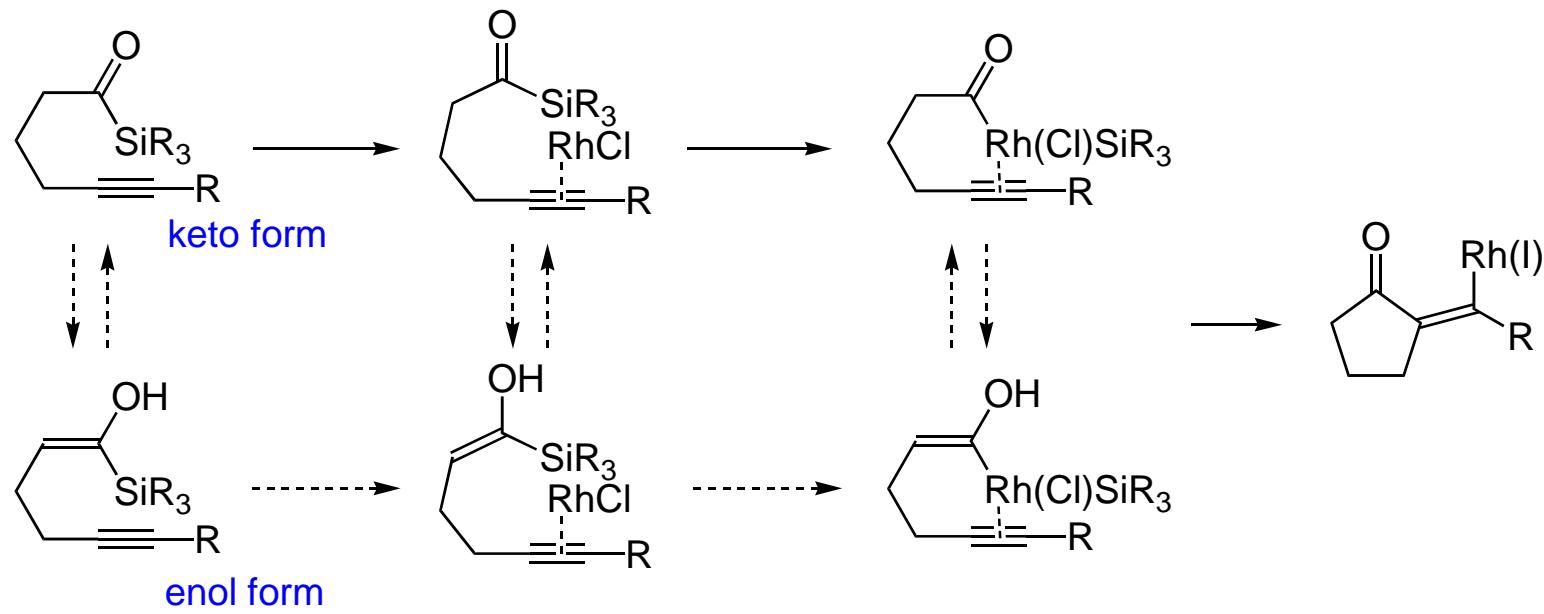
R'	<b>SiR<sub>3</sub></b>	temp.	time	yield
Ph	SiMe <sub>2</sub> Ph	70 °C	12 h	82%
		100 °C	5 h	80%
(CH <sub>2</sub> ) <sub>2</sub> Ph	SiMe <sub>2</sub> Ph	70 °C	12 h	77%
Ph	SiMe <sub>3</sub>	70 °C	24 h	77%
	SiMe <sub>3</sub>	100 °C	3 h	81%

*Chem. Lett.* **2001**, 1210.





## Acylsilane or Vinylsilane ?



pKa

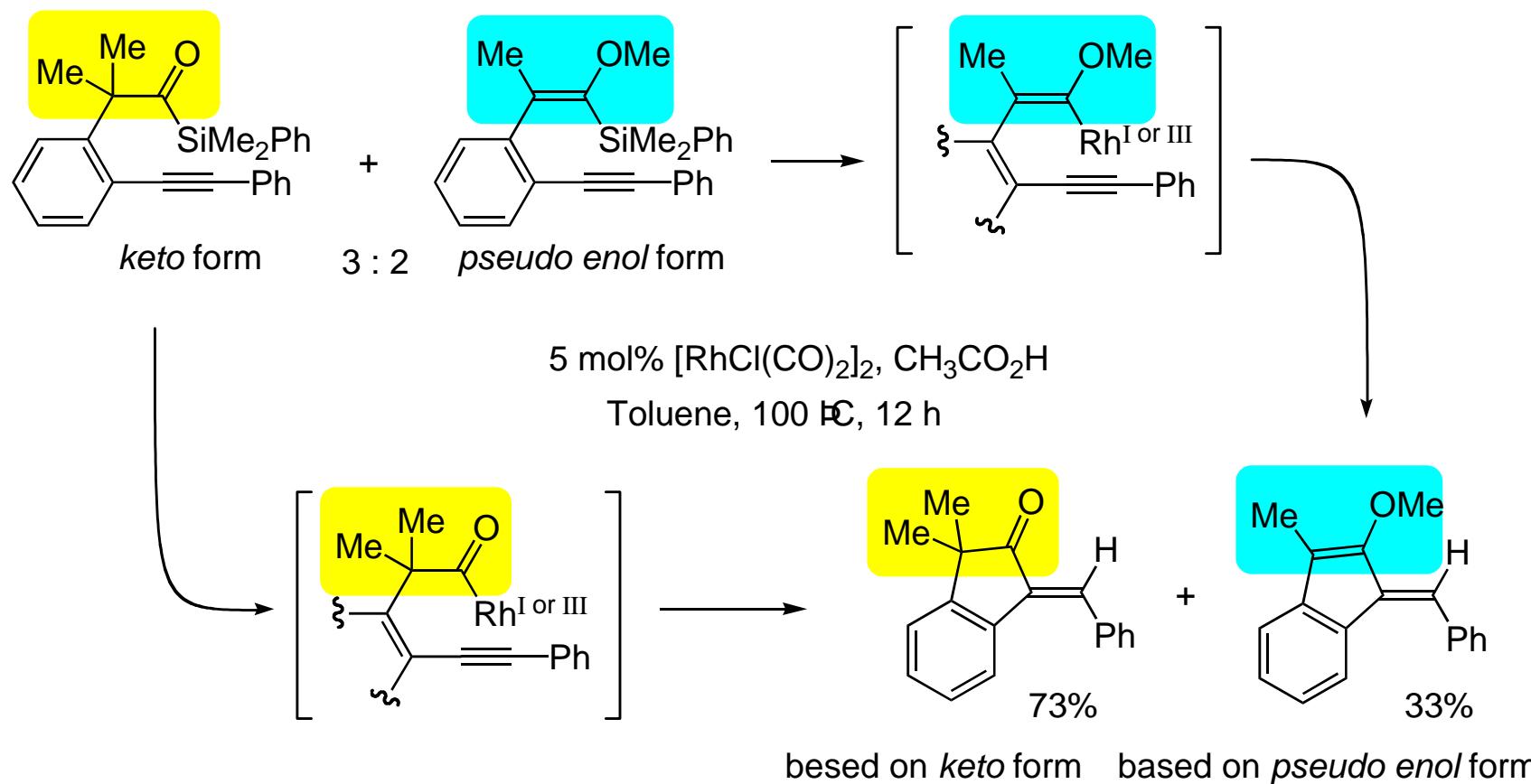
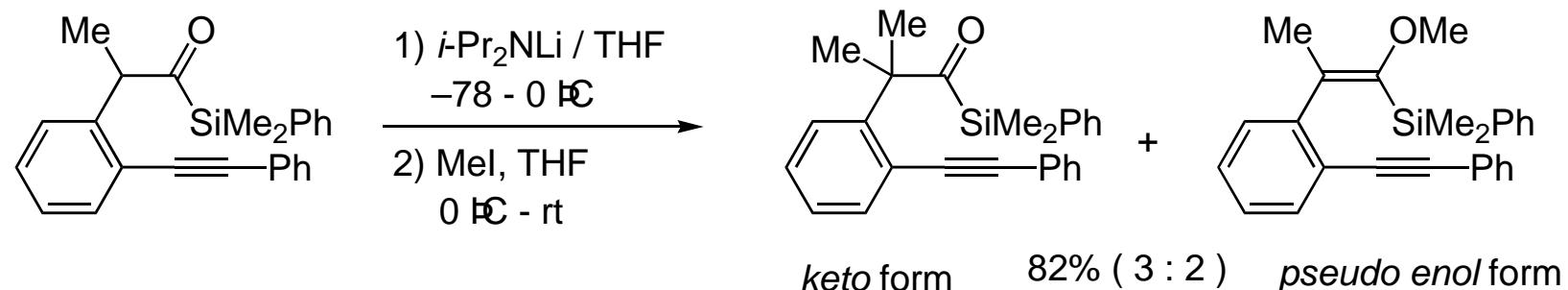
R = SiMe<sub>3</sub> 16.4

R = CH<sub>3</sub> 19.3

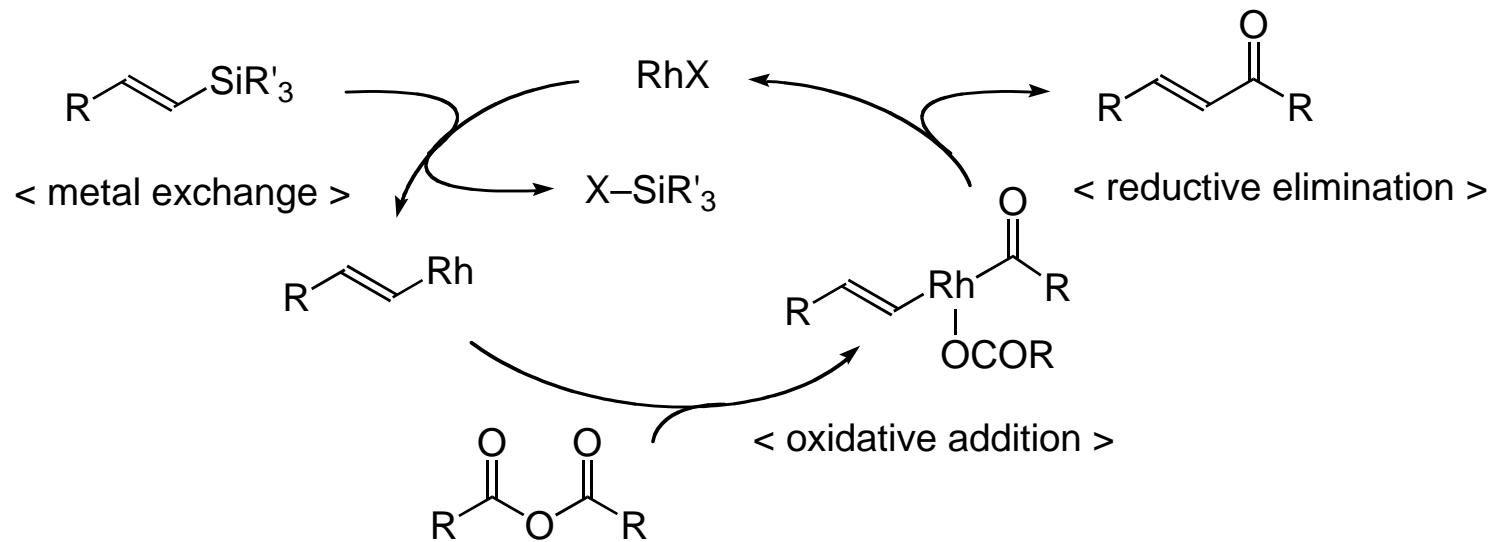
i-C<sub>3</sub>H<sub>7</sub>OH 16.5

A. J. Kresge, J. B. Tobin

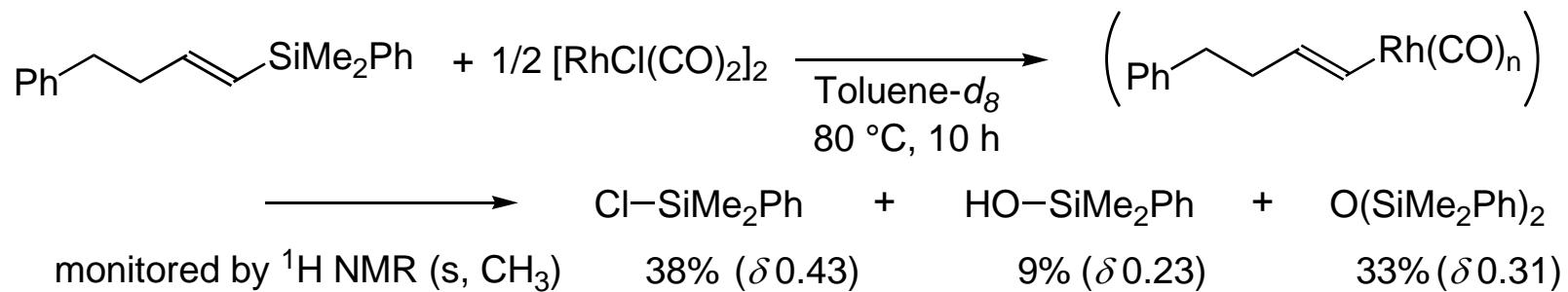
*J. Am. Chem. Soc.* **1990**, *112*, 2805.



## Catalytic Acylation of Vinylsilanes

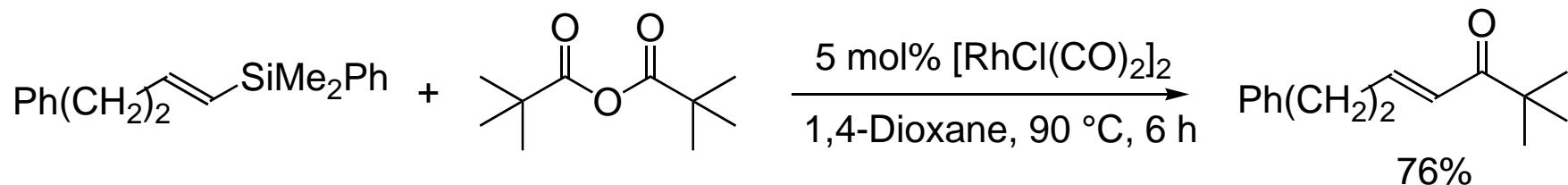


### Stoichiometric Reaction

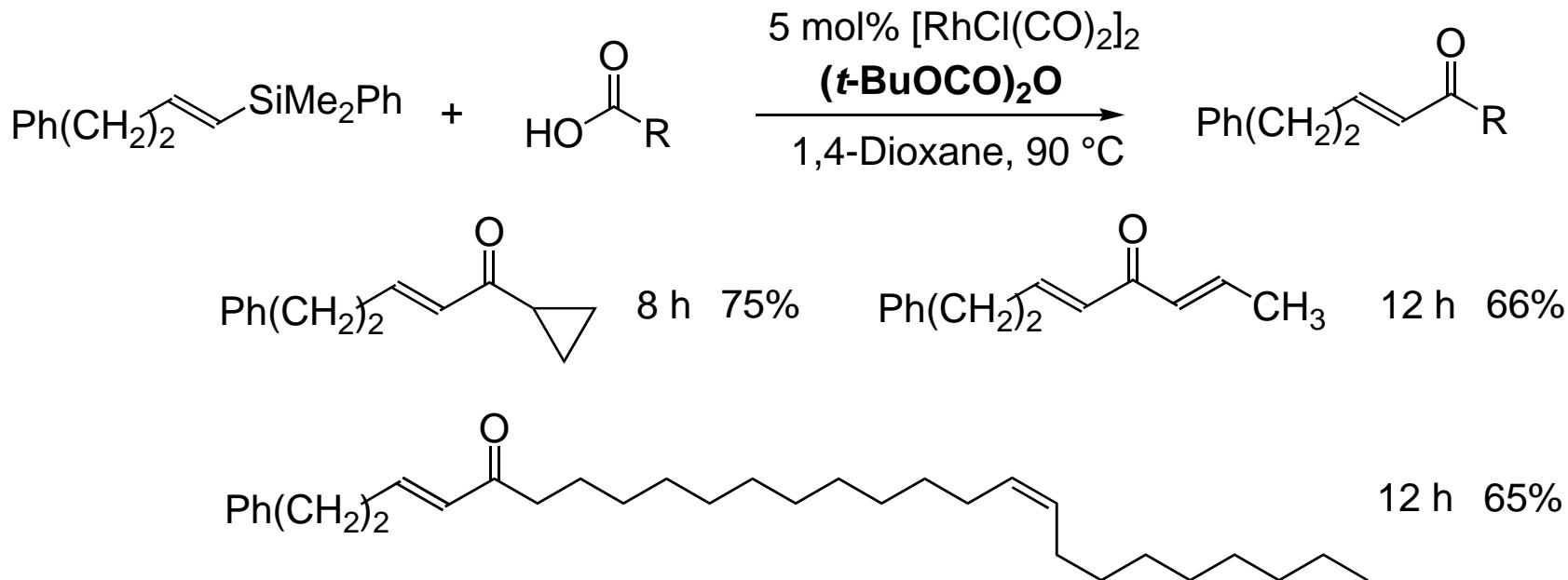


## Catalytic Acylation of Vinylsilanes

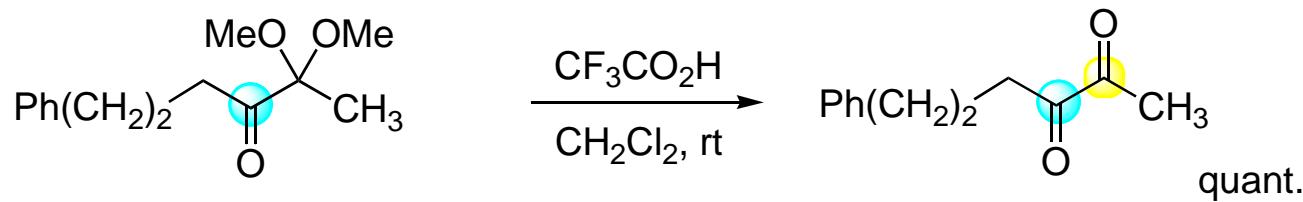
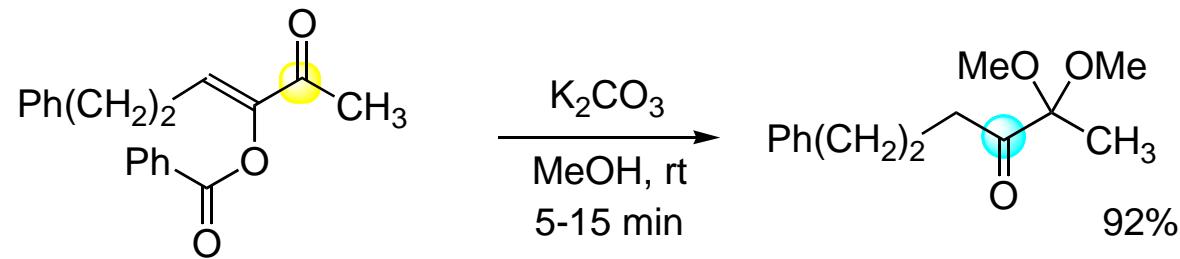
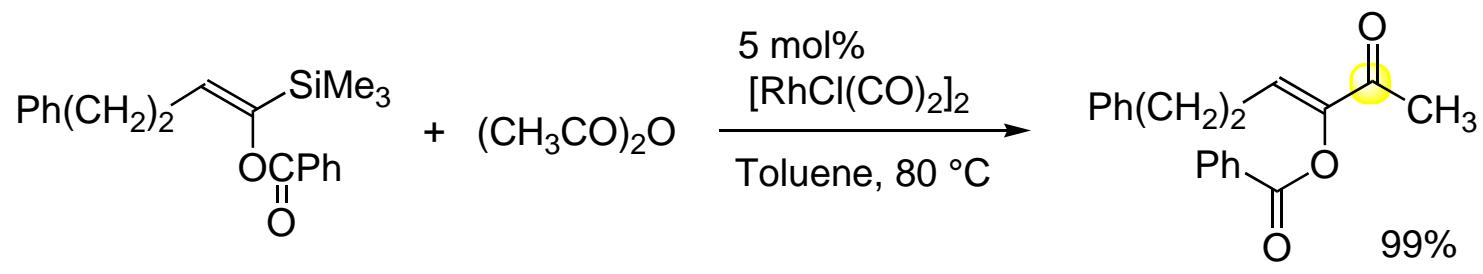
with acid anhydrides



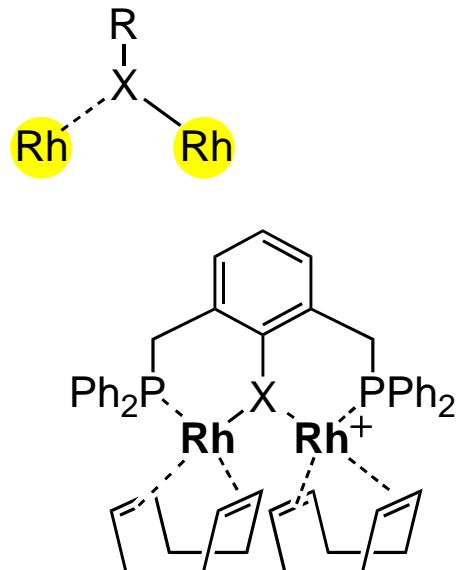
with carboxylic acids



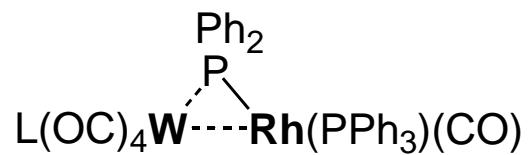
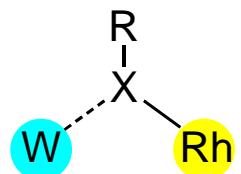
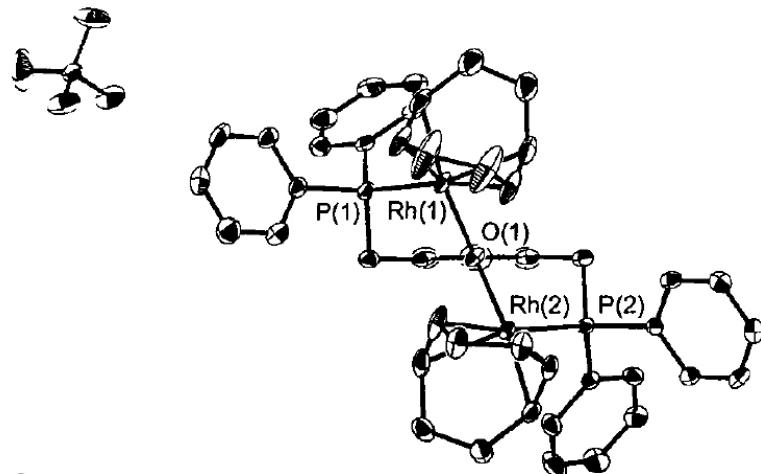
## Preparation of Unsymmetrical $\alpha$ -Diketones



## Hetroatom-Bridged Bimetallic Complexes

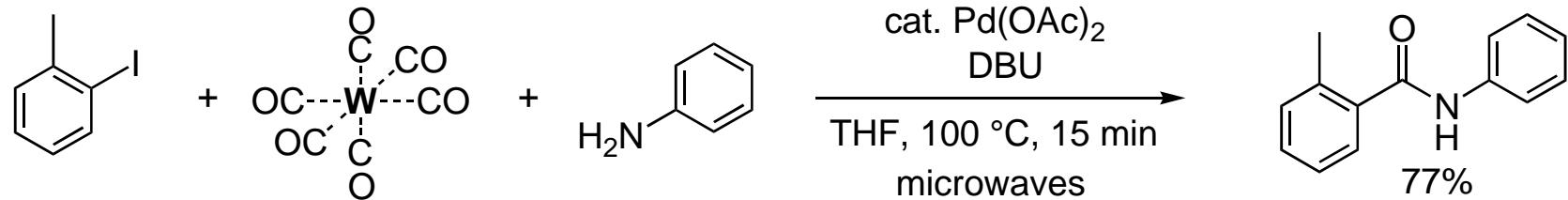
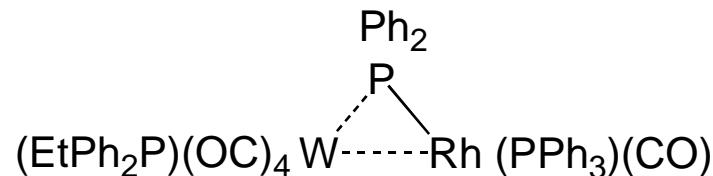
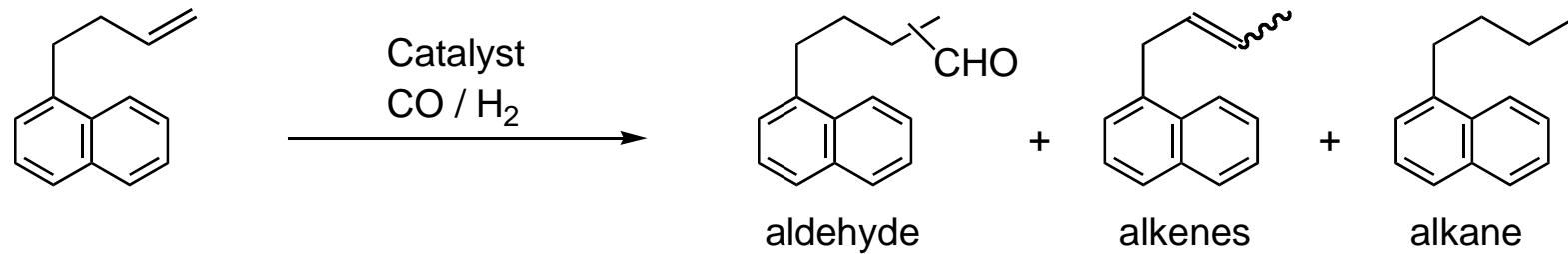


X = O, X = S

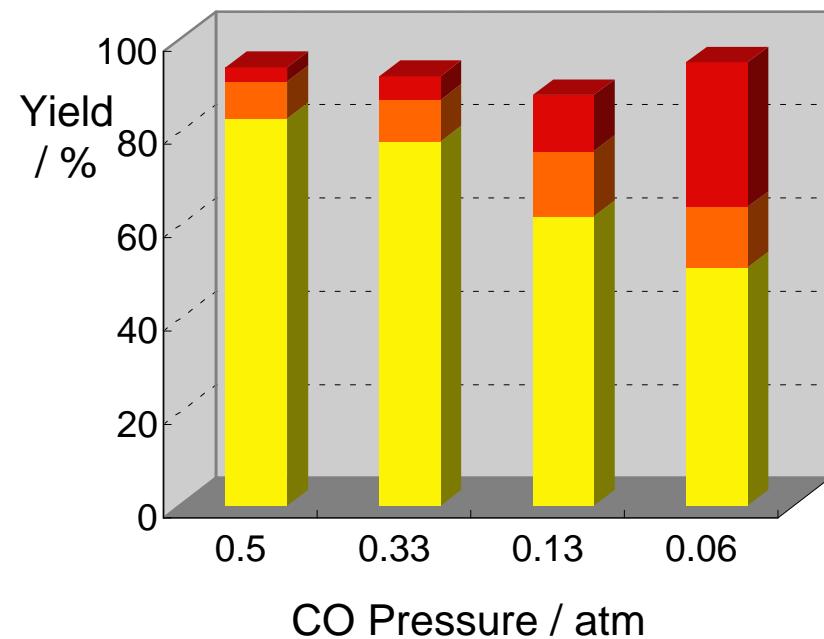
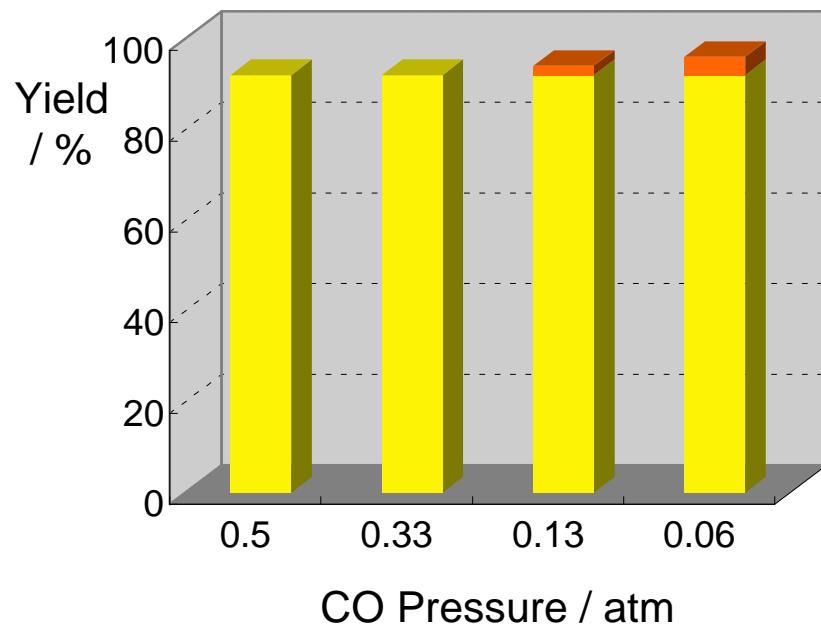
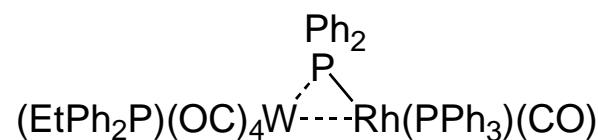
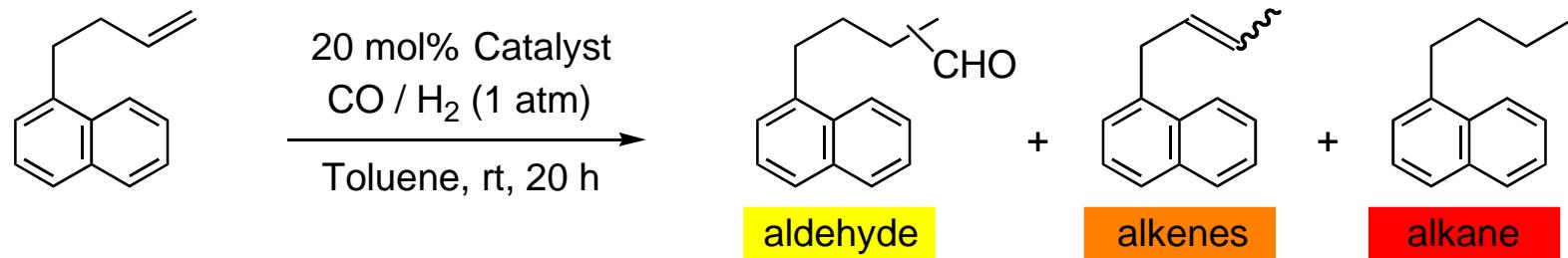


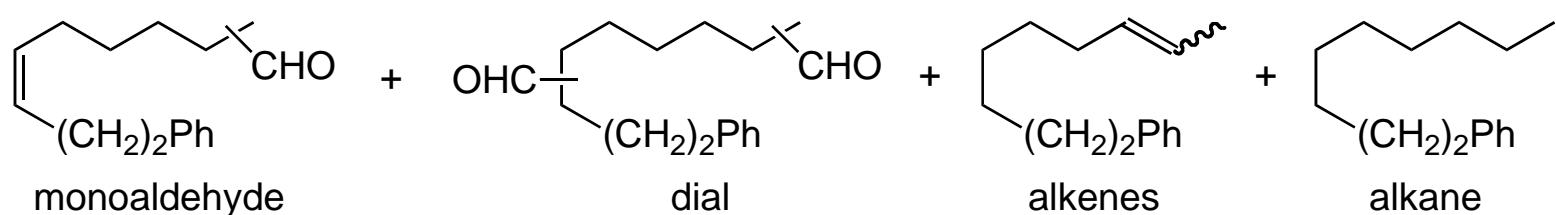
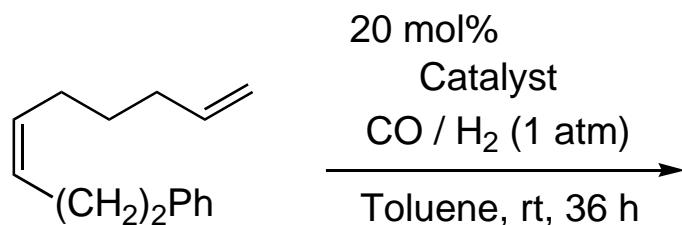
P. M. Shulman et al, *Organometallics*, 1987, 6, 101.

# Hydroformylation of Alkenes



N. F. K. Kaiser, A. Hallberg, K. Larhed, *J. Comb. Chem.* **2002**, 4, 109.

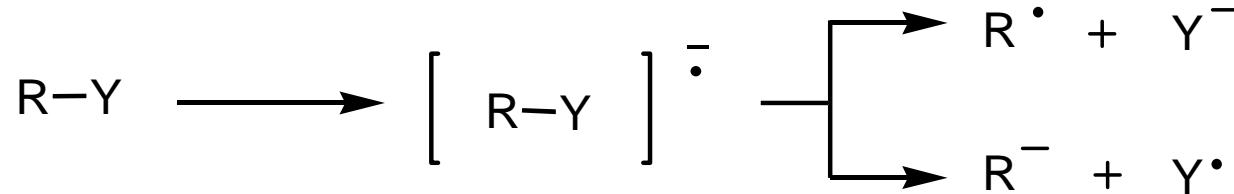
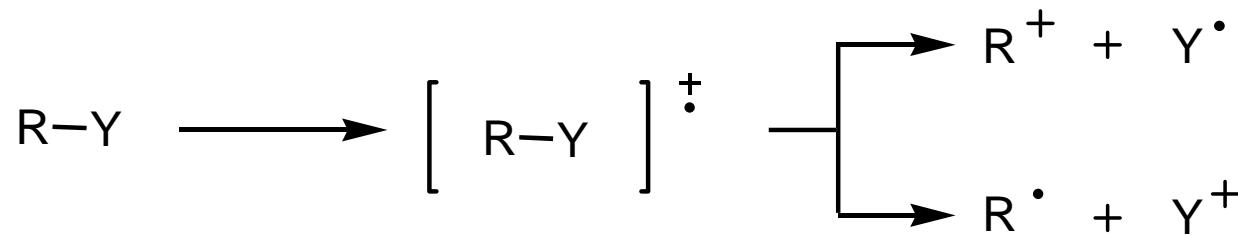
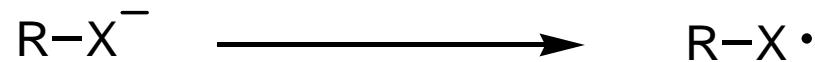




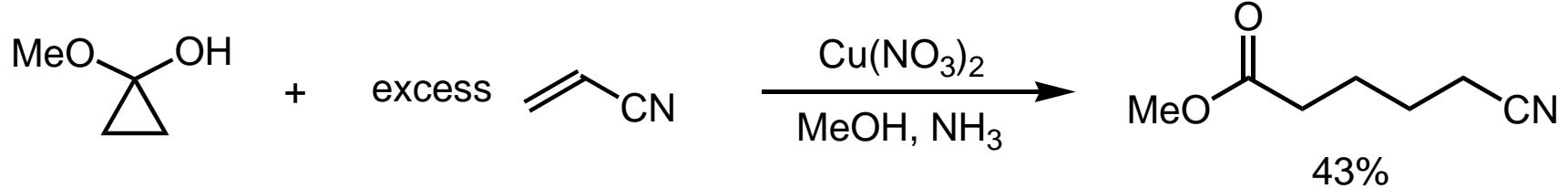
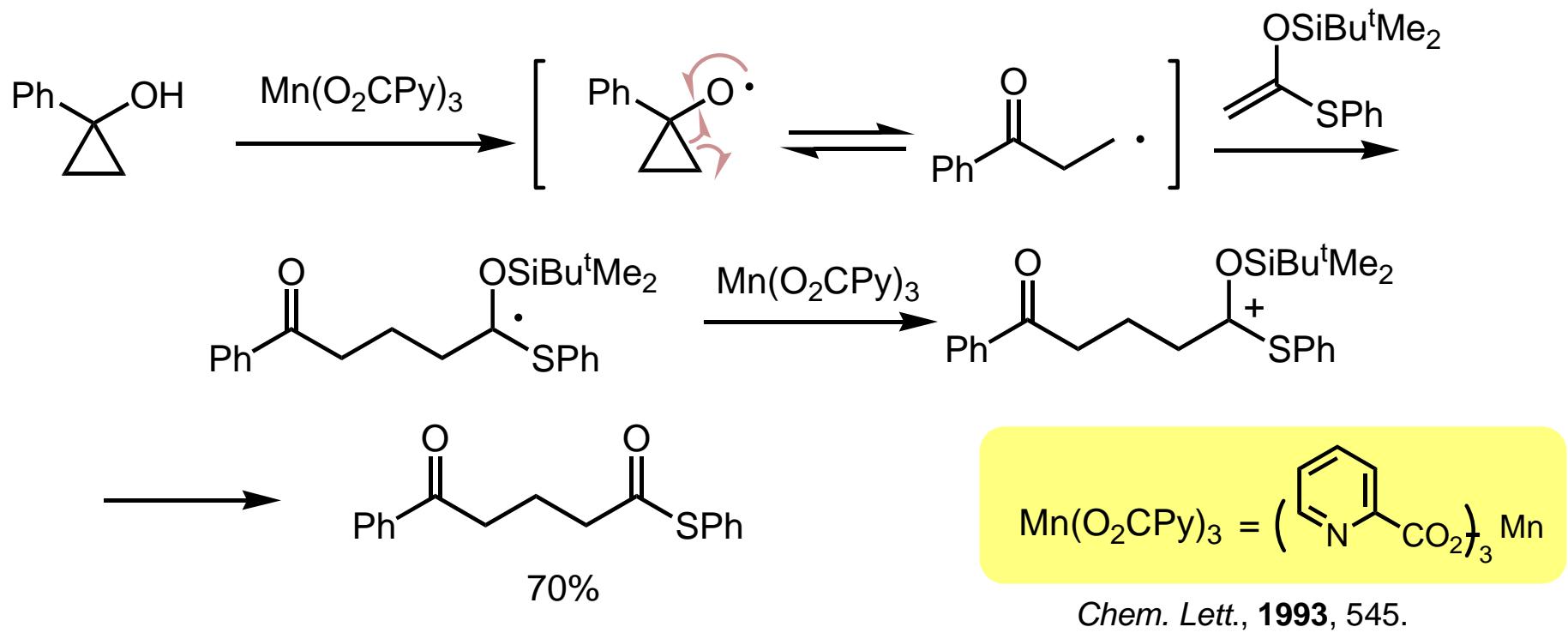
	monoaldehyde	dial	alkenes	alkane
$(EtPh_2P)(OC)_4W\dashdots Rh(PPh_3)(CO)$	89% ( <i>n</i> : <i>iso</i> = 2.5 : 1)	0%	4%	0%
$RhH(CO)(PPh_3)_3$	60% ( <i>n</i> : <i>iso</i> = 2.5 : 1)	<12%	14%	10%

# Redox-Catalysts for Generation of Radical Species

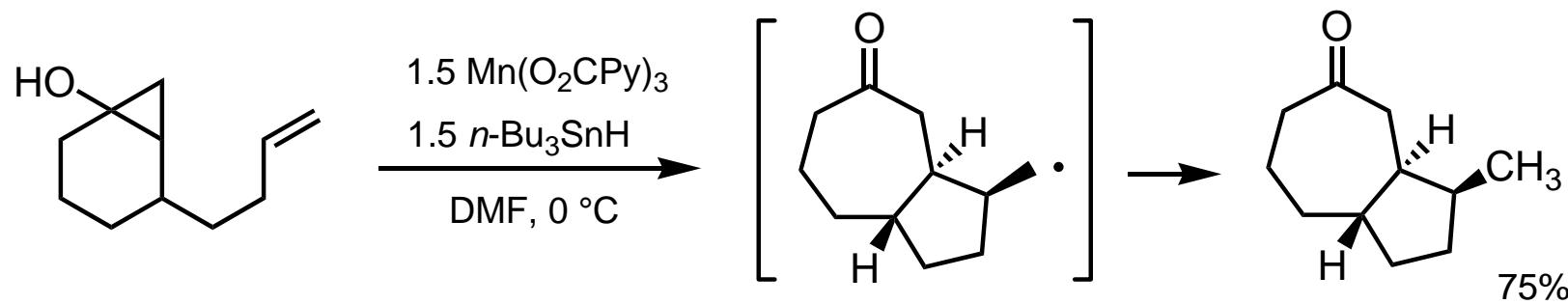
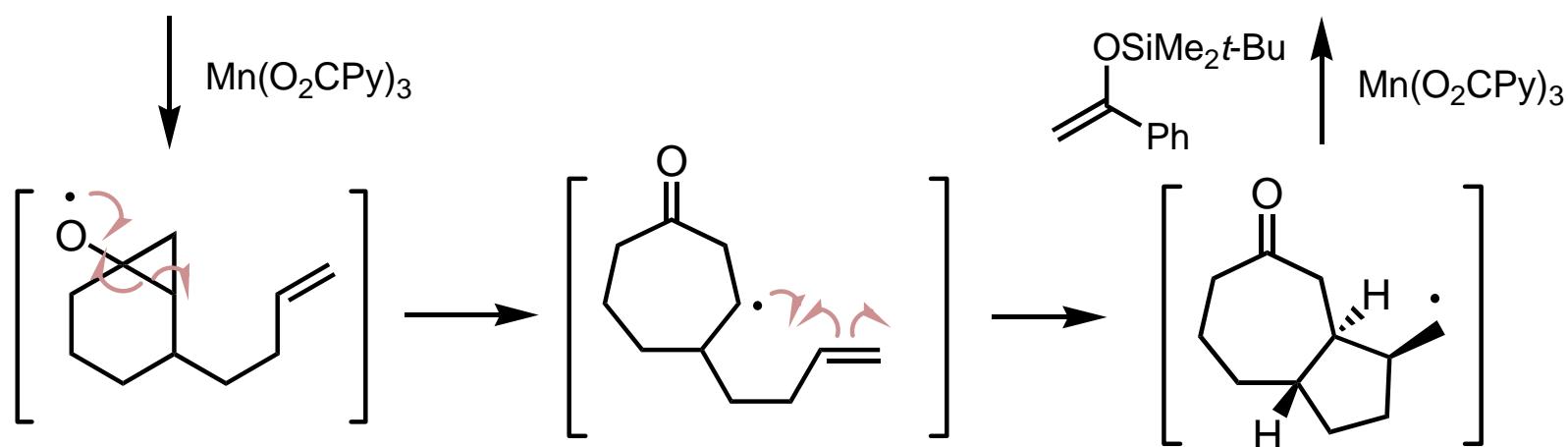
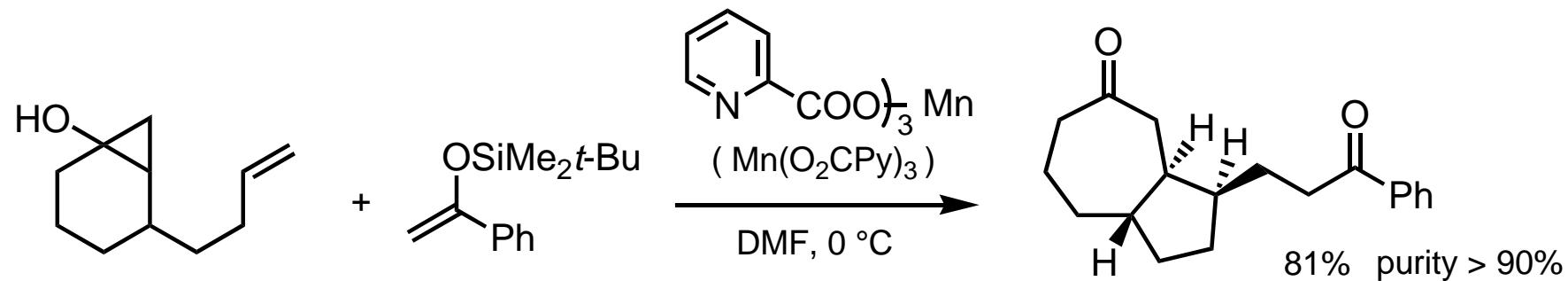
1989 - present



## Generation of $\beta$ -Keto Radicals: Intermolecular Addition to Olefins



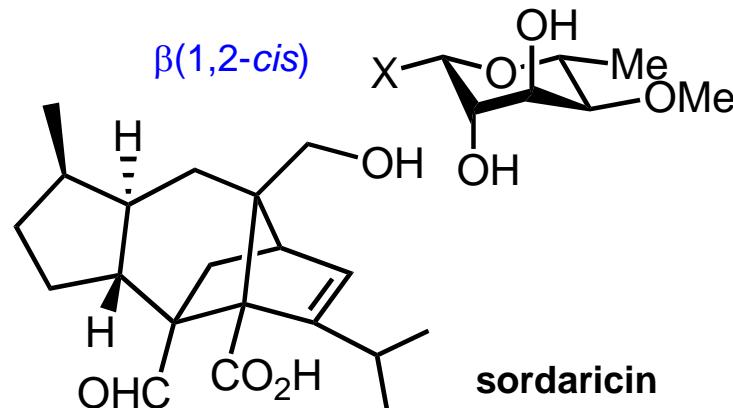
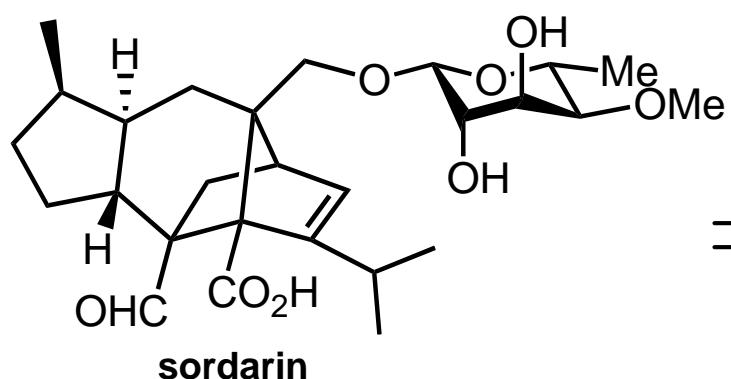
S. E. Schaafsma et al., *Tetrahedron Lett.*, 1973, 827.



*Chem. Lett.*, 1994, 1697.

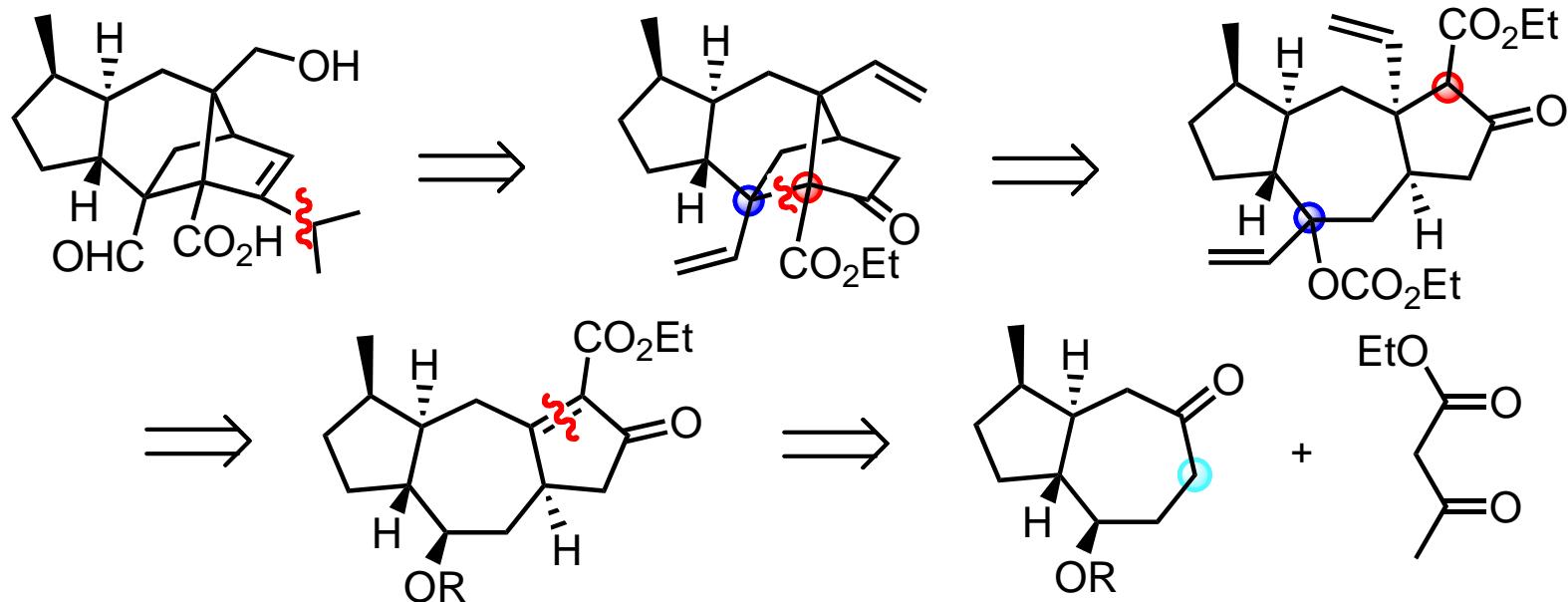
## Synthesis of (-)-Sordarin

Synthesis of (-)-Sordarin: see *J. Am. Chem. Soc.*, **128**, 6931 (2006).

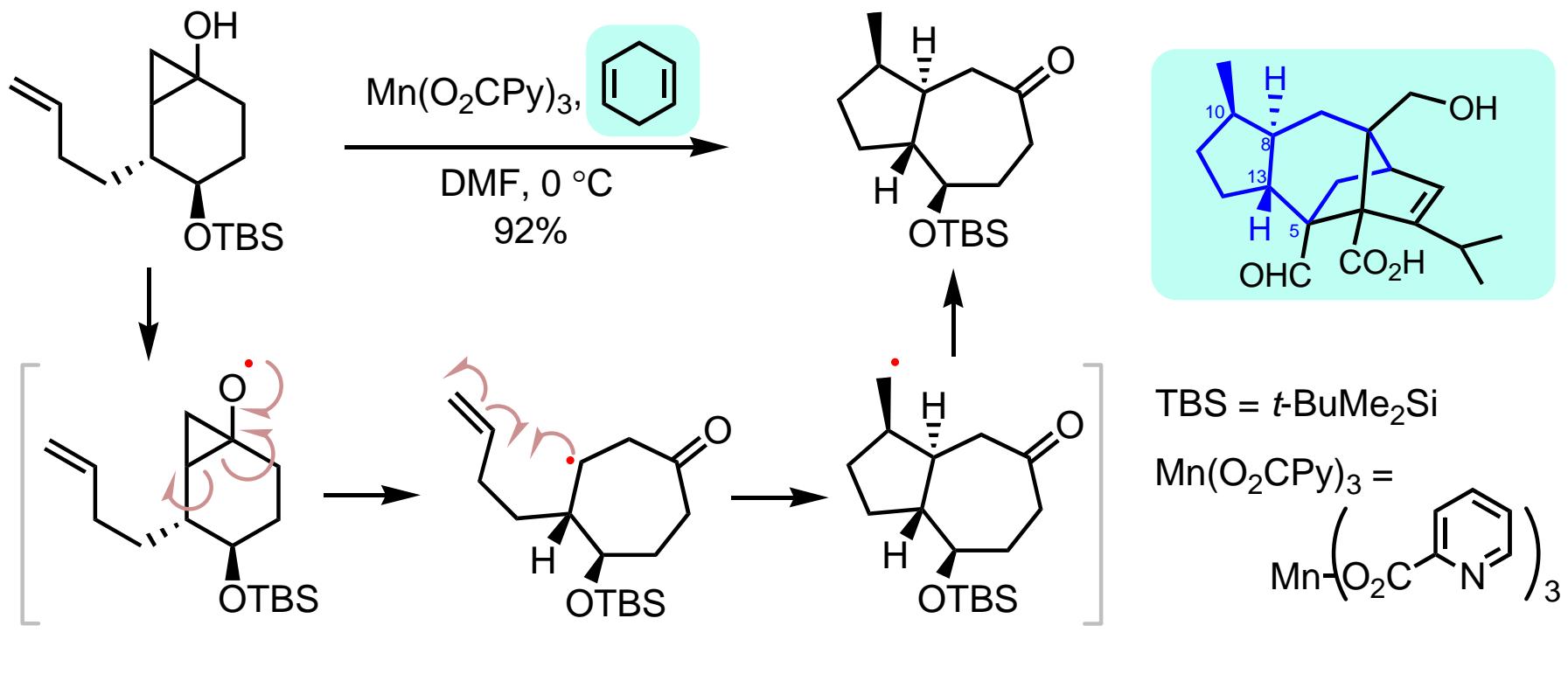


D. Hauser, et al., *Helv. Chim. Acta*, **54**, 1178 (1971).

selective inhibitor of fungal protein synthesis: M. C. Justice, et al., *J. Biol. Chem.*, **273**, 3148 (1998).

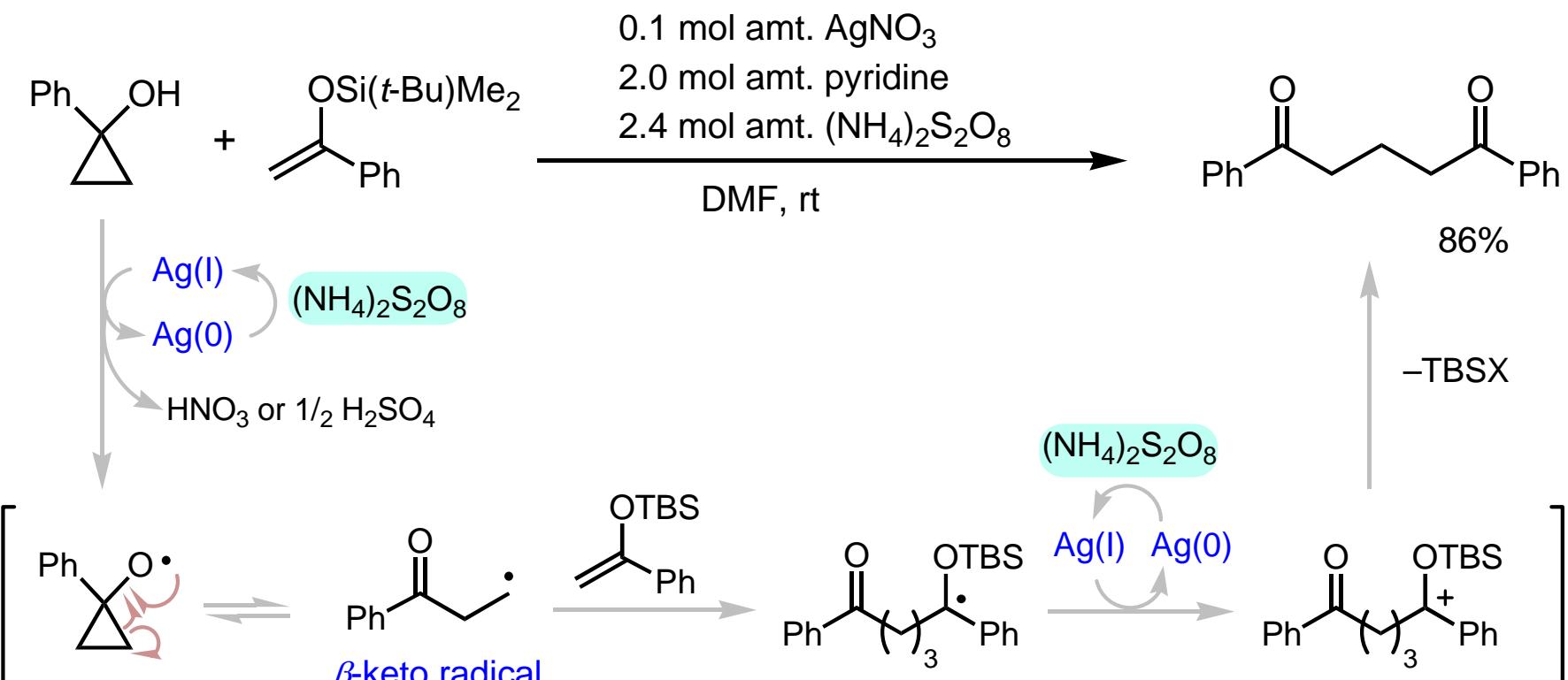


## Preparation of Bicyclo[5.3.0]decan-3-one Derivative

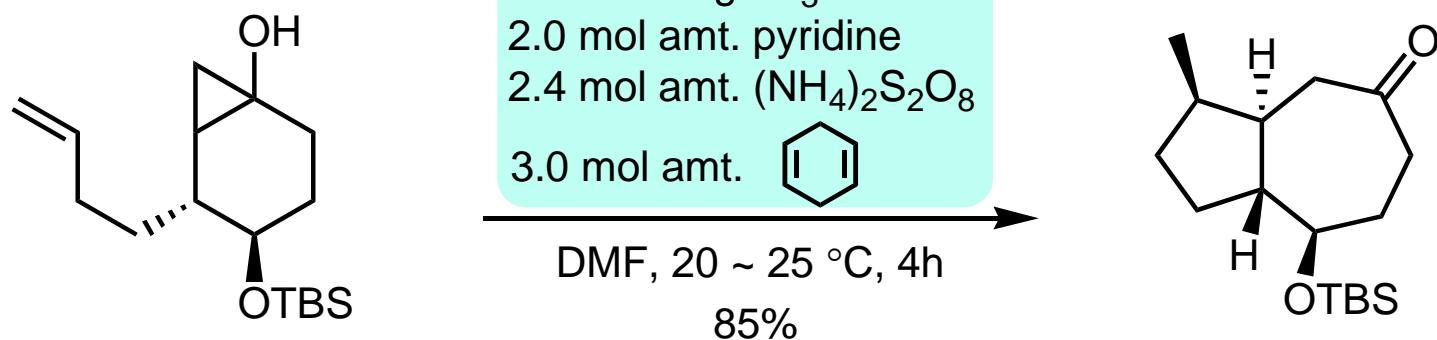
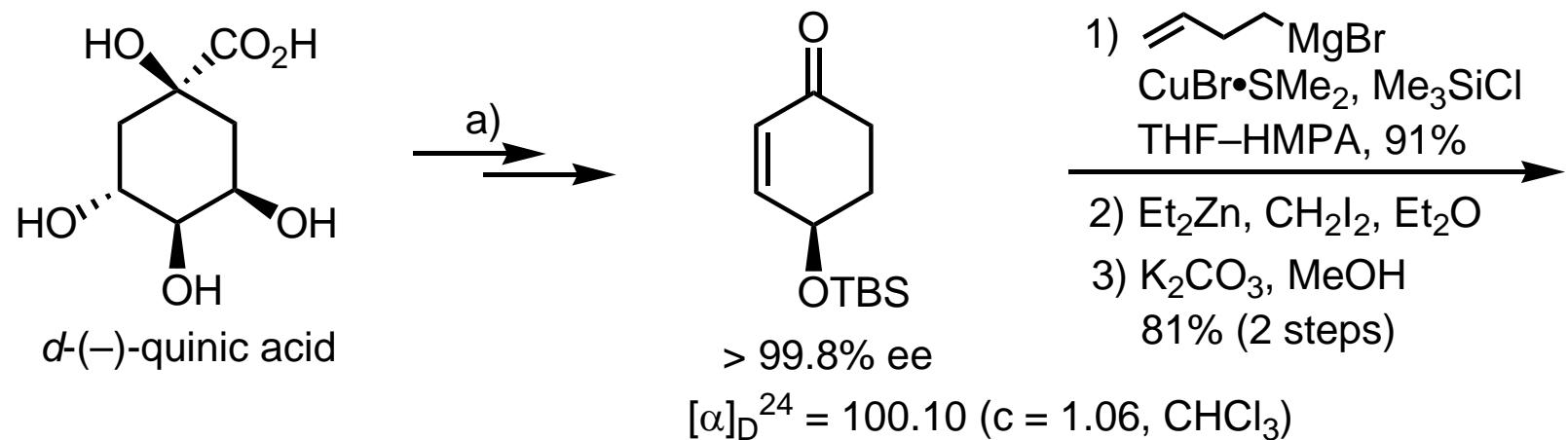


*Bull. Chem. Soc. Jpn.*, **72**, 85 (1999).

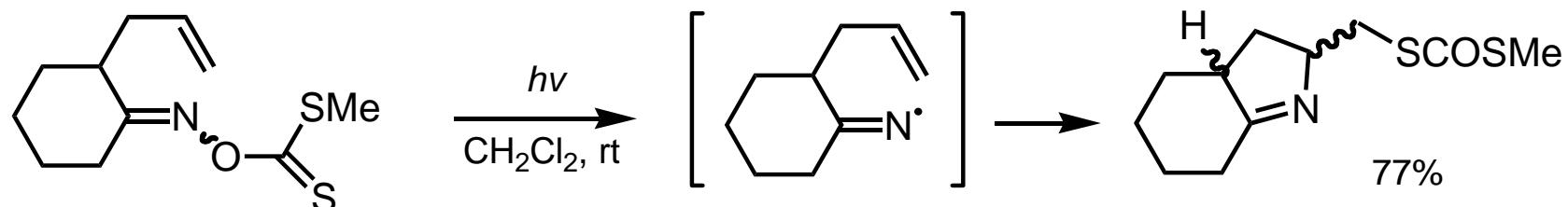
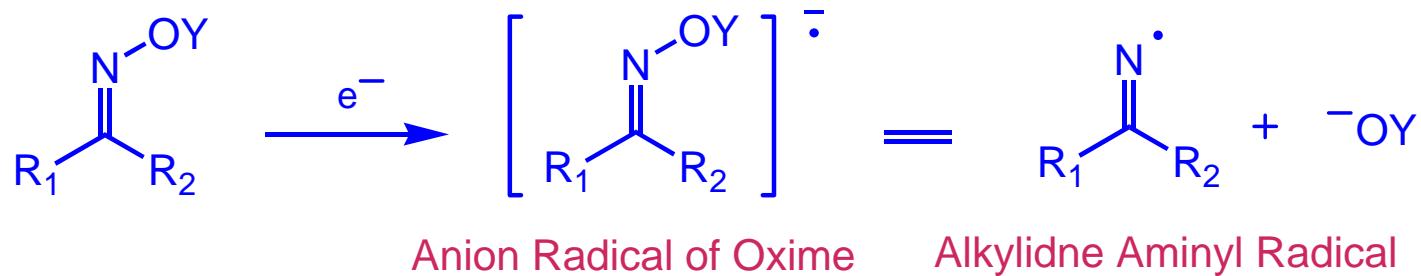
## Ag(I)-Pyridine Catalytic System



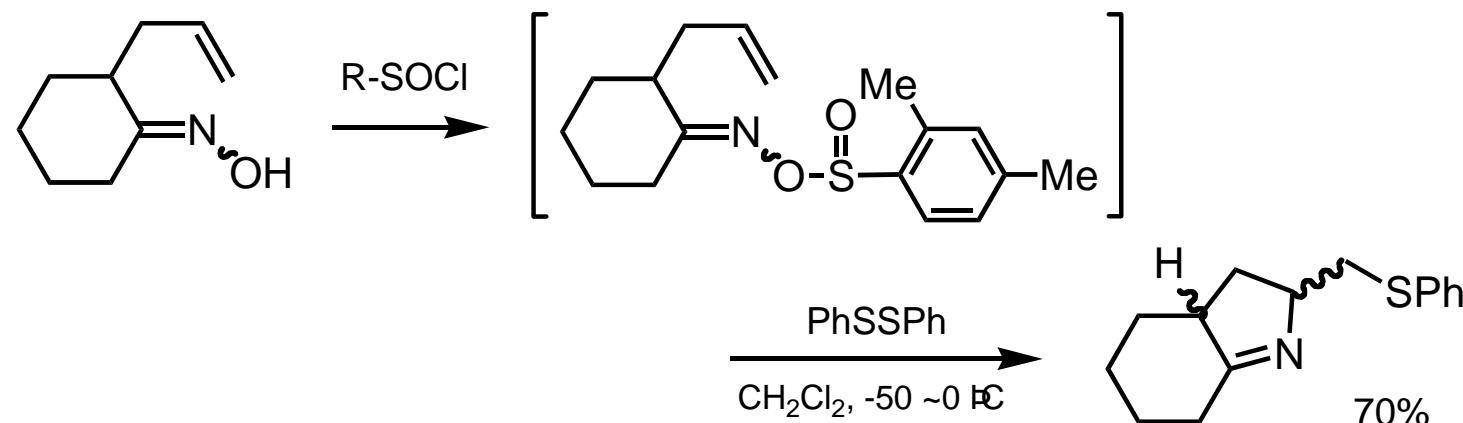
*Chem. Lett.*, **35**, 18 (2006).



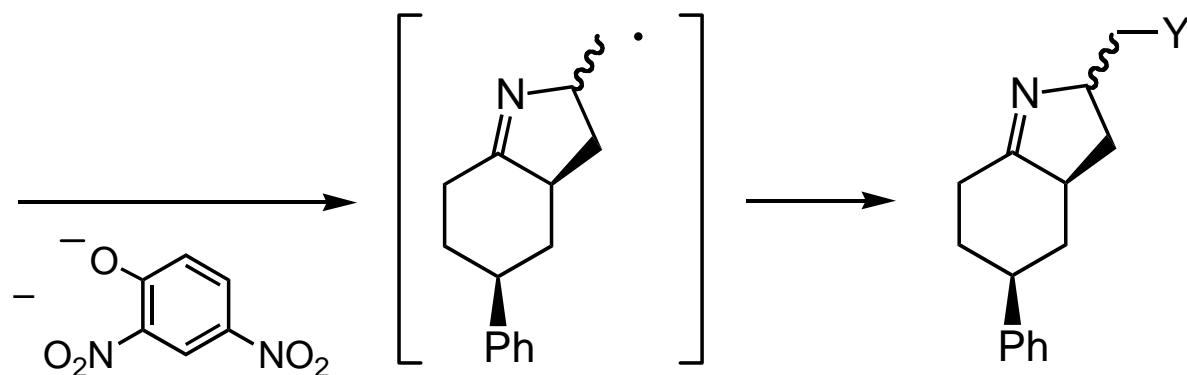
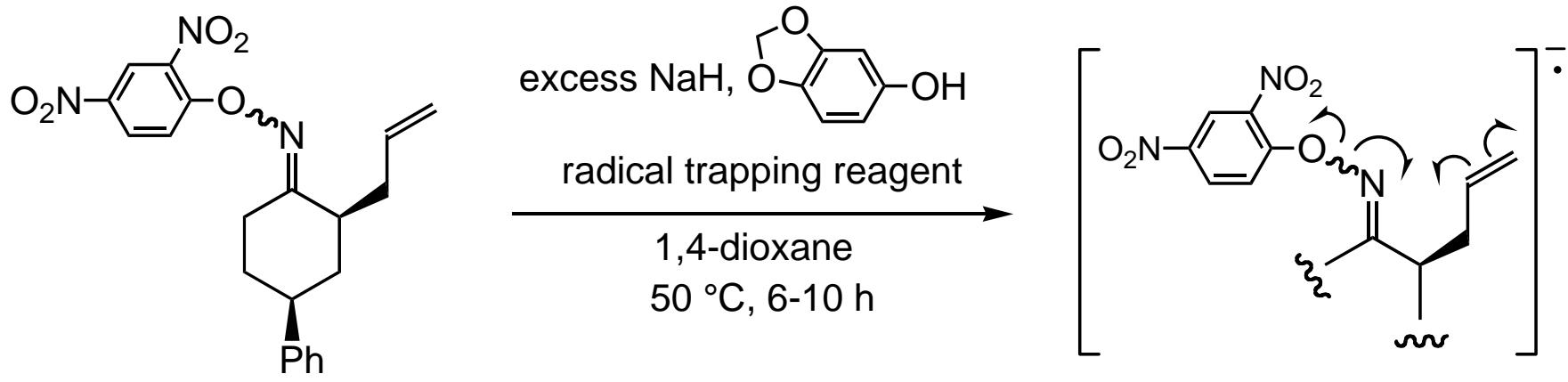
a) C. D. Maycock, et al., *J. Chem. Soc., Perkin Trans. I*, 2001, 166.



F. Gagosz and S. Z. Zard, *Synlett*, **1999**, 1978.



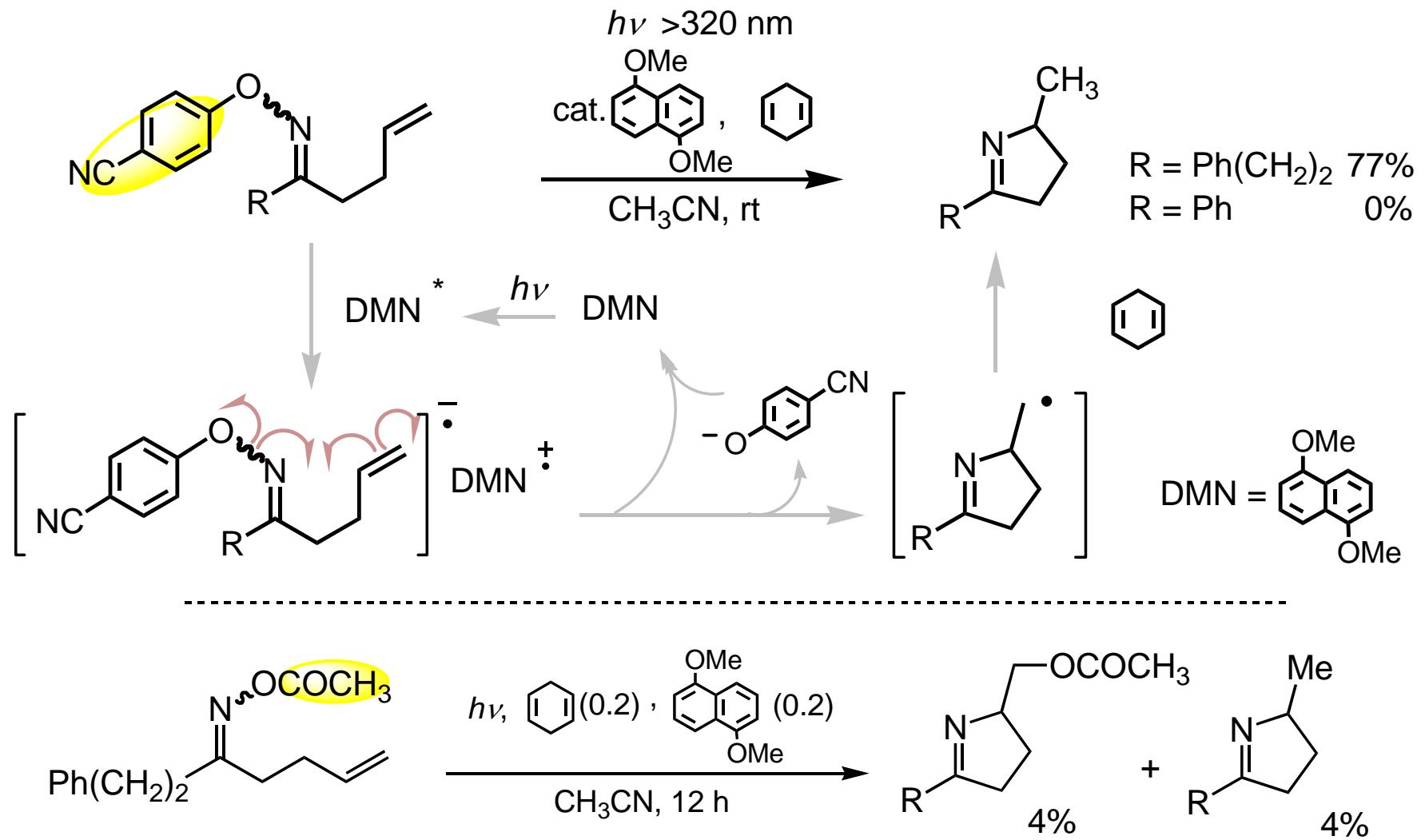
S. M. Weinreb et al., *Org. Lett.*, **1**, 637 (1999).

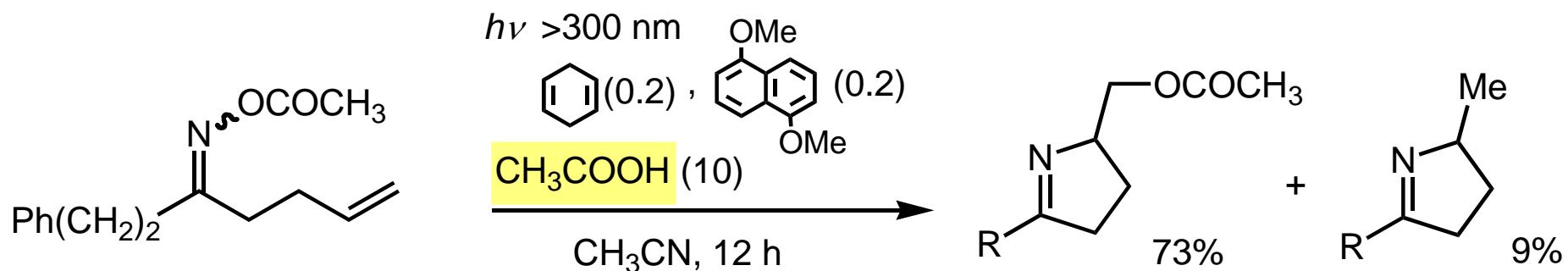


	91% (Y = H)
	75% (Y = Cl)
	70% (Y = SPh)
	69% (Y = SePh)

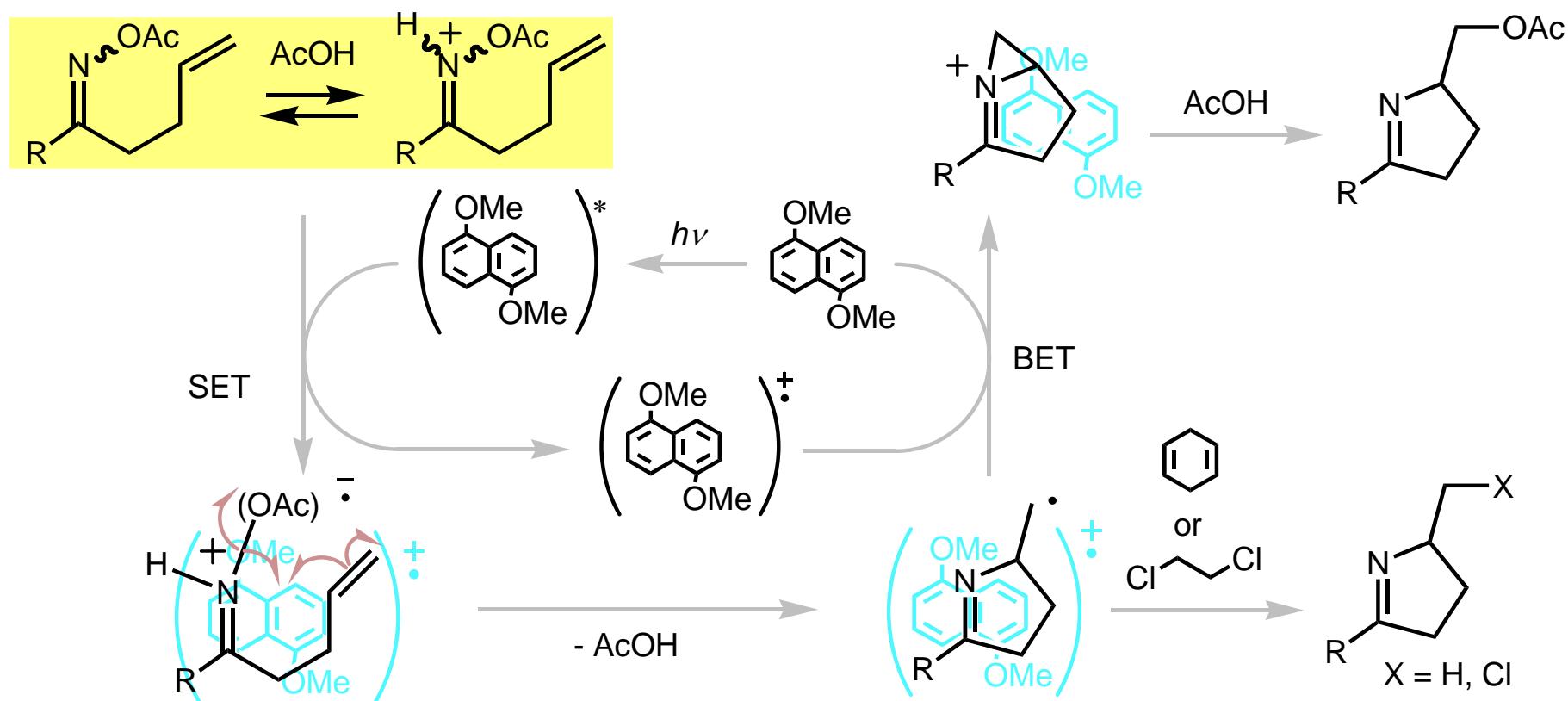
*Chem. Lett.*, 1998, 1261.

## Photochemical Electron Transfer

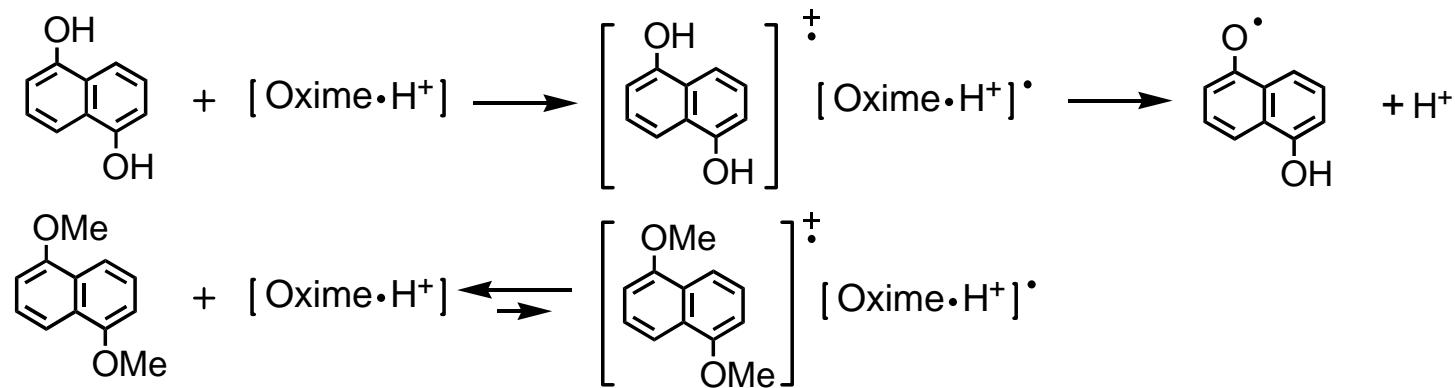
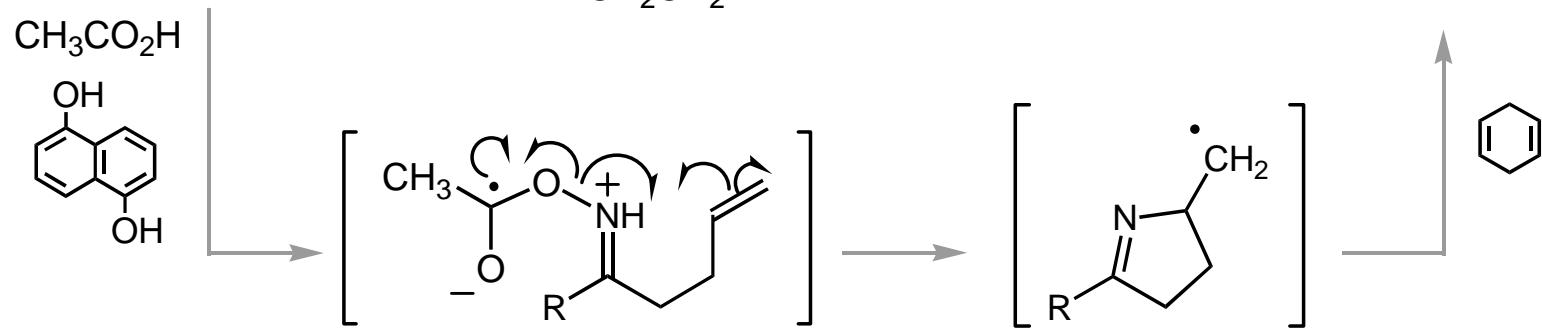
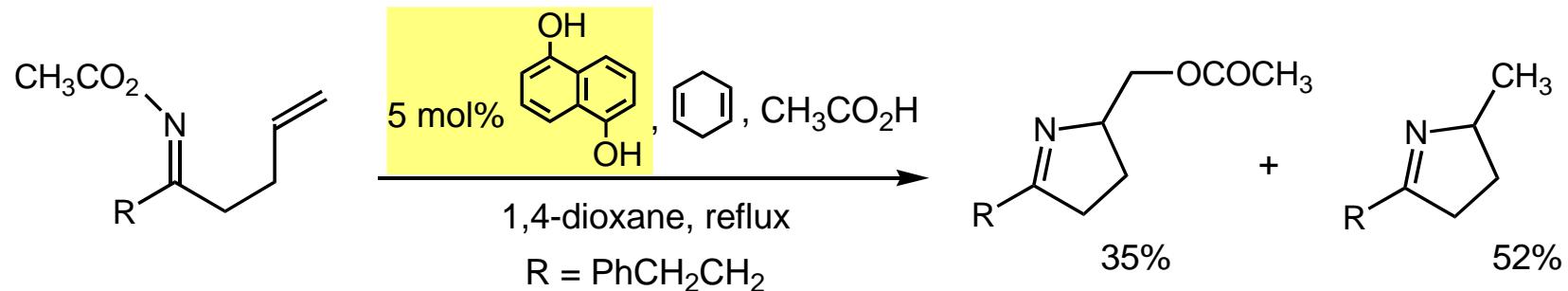


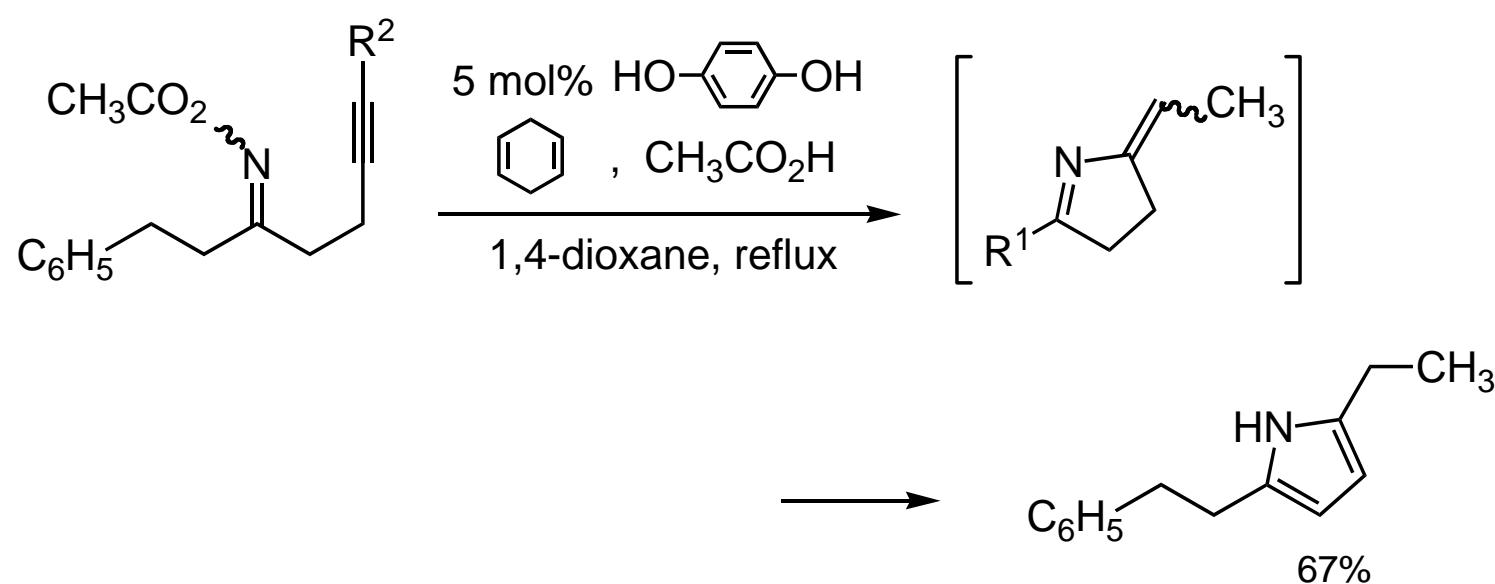
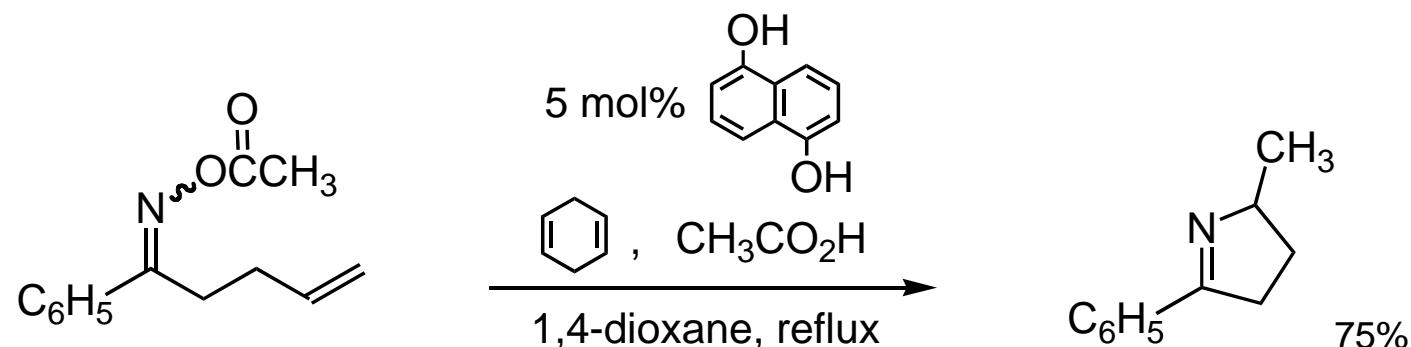


Tetrahedron Lett., 46, 2373 (2005).

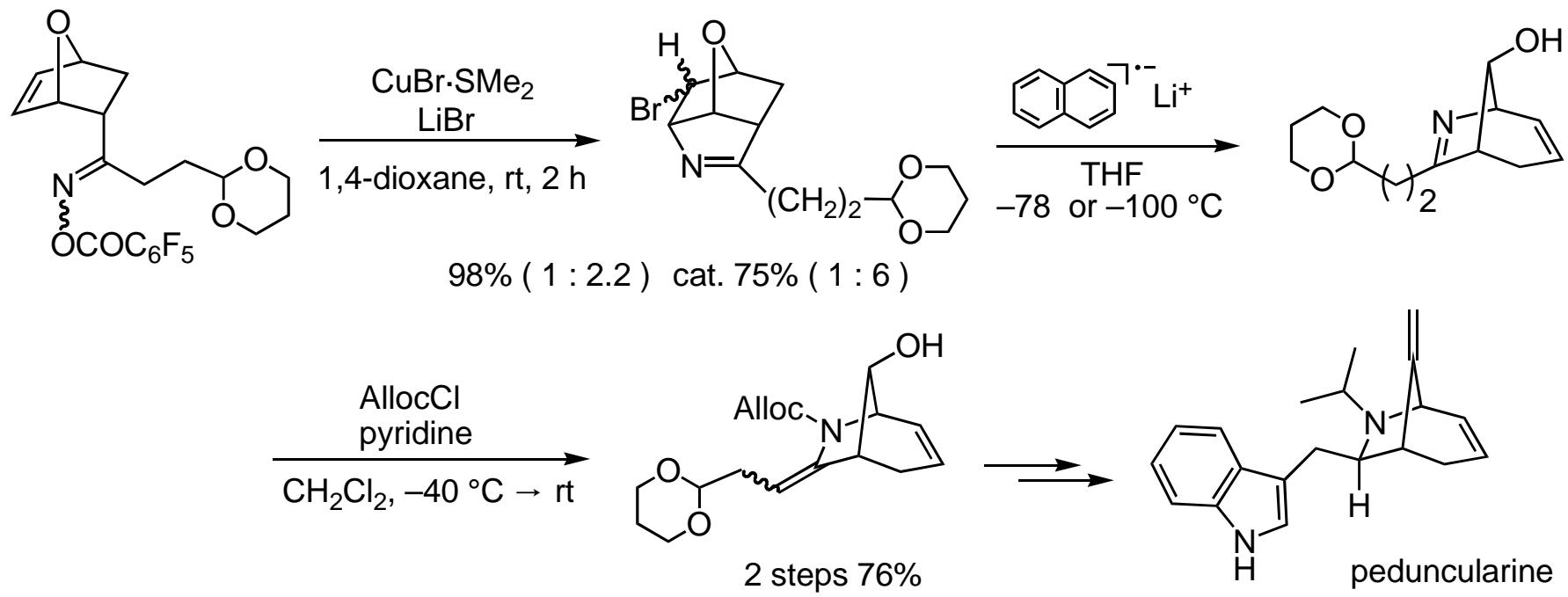
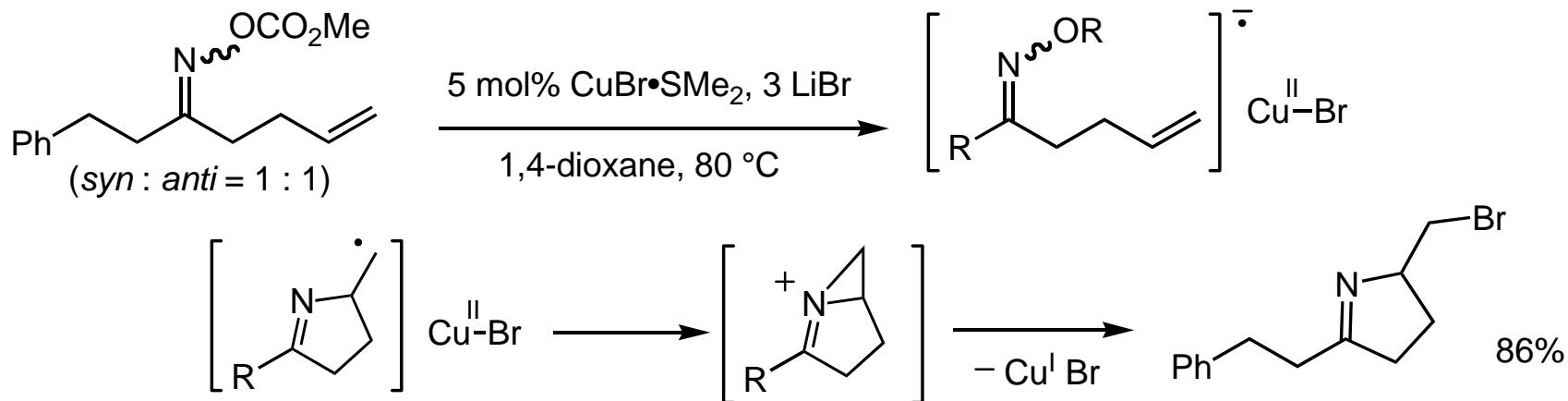


## Non-Photochemical Electron Transfer Catalysts

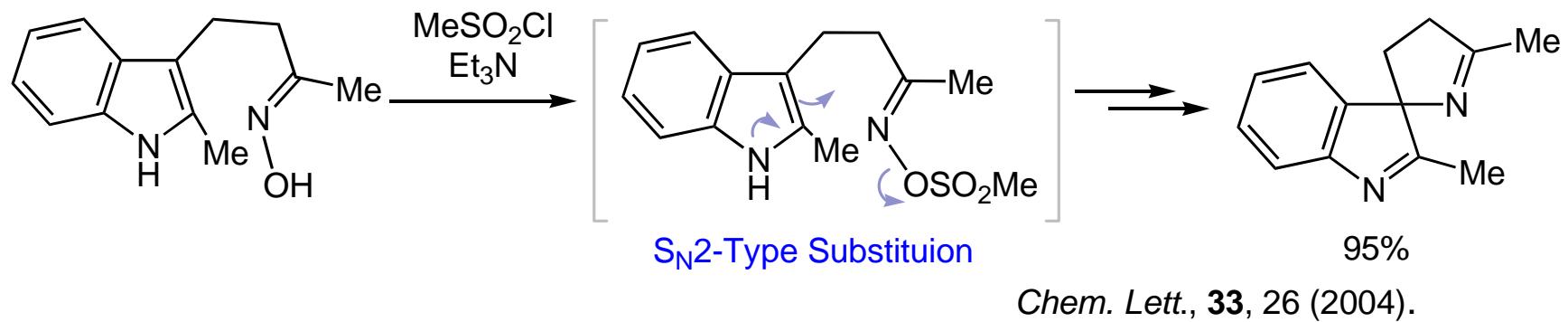
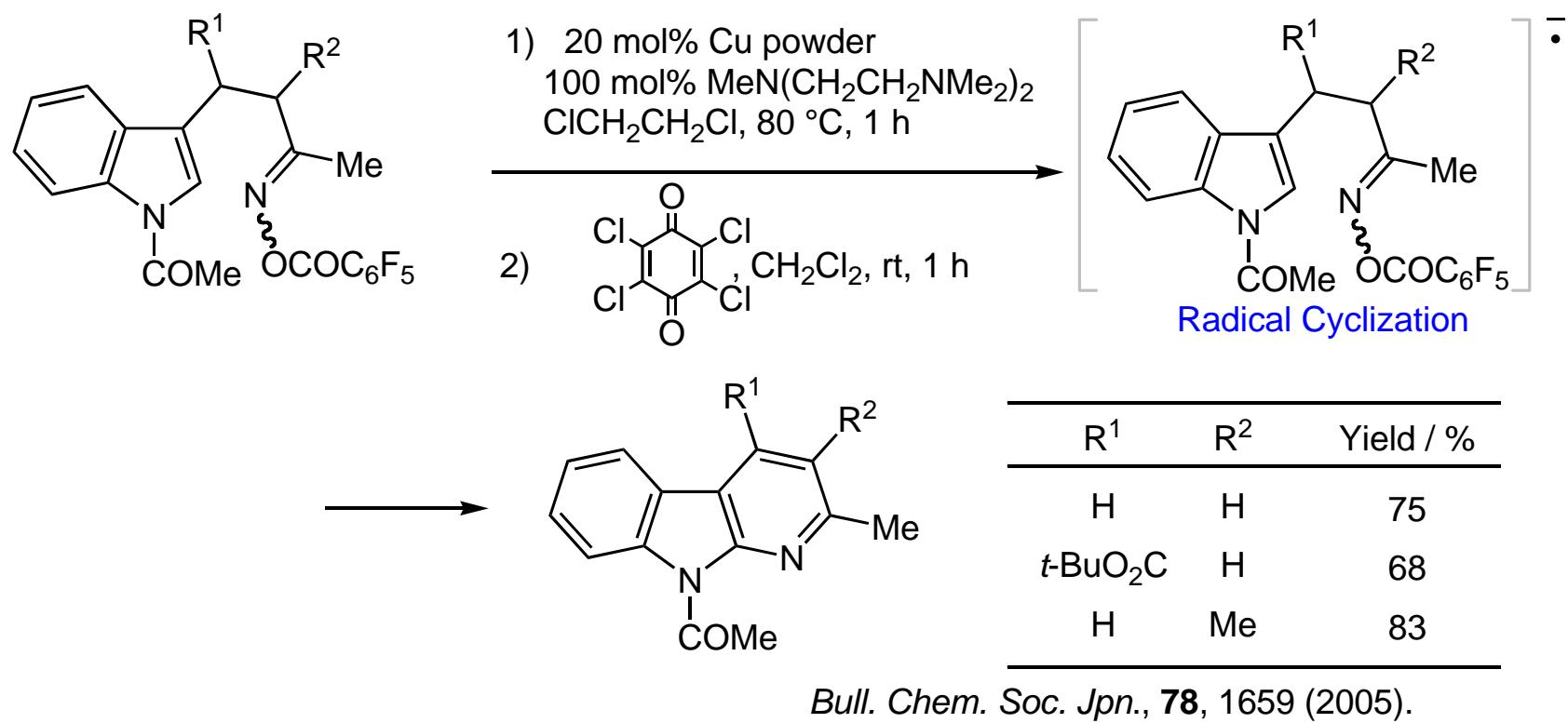




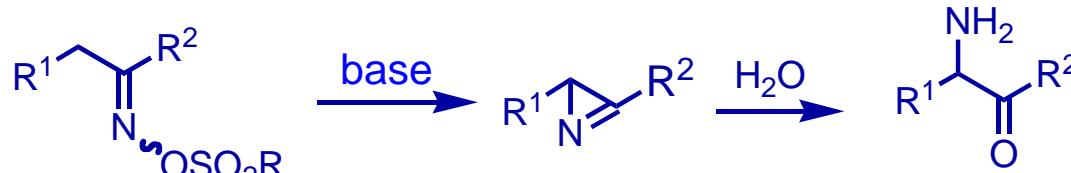
## Preparation of 3,4-Dihydro-2*H*-pyrroles with Cu(I)-Catalyst



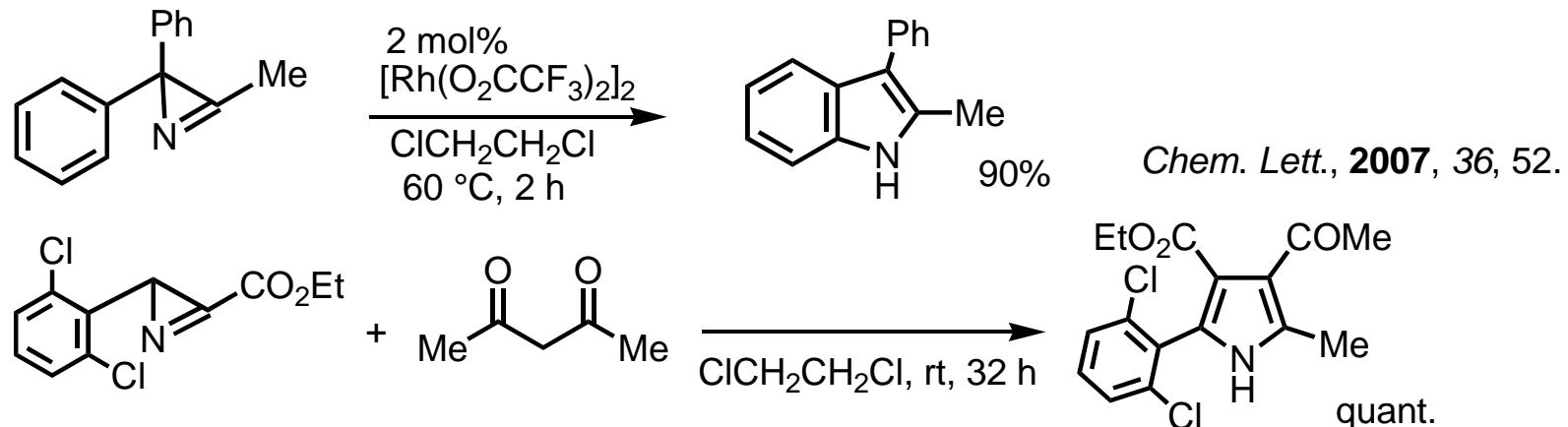
*Bull. Chem. Soc. Jpn.*, **79**, 1552 (2006).



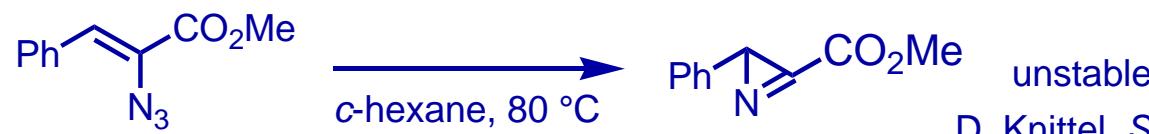
Oxime  $\longrightarrow$  2*H*-Azirine



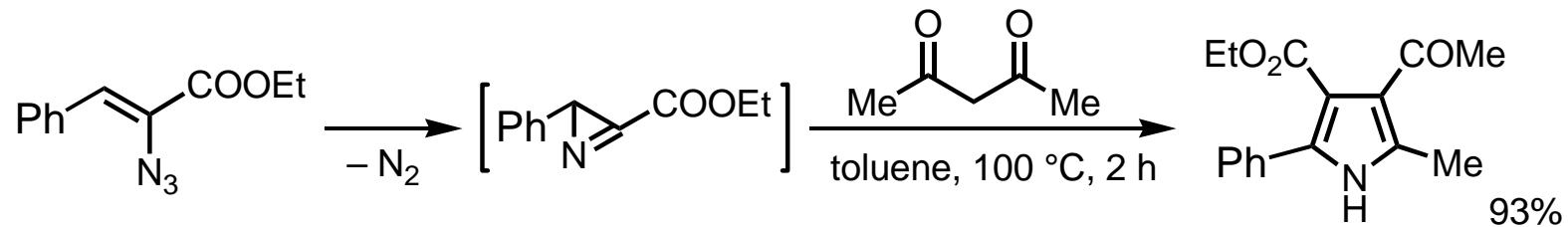
P. W. Neber, et al. *Ann.*, **449**, 109 (1926).



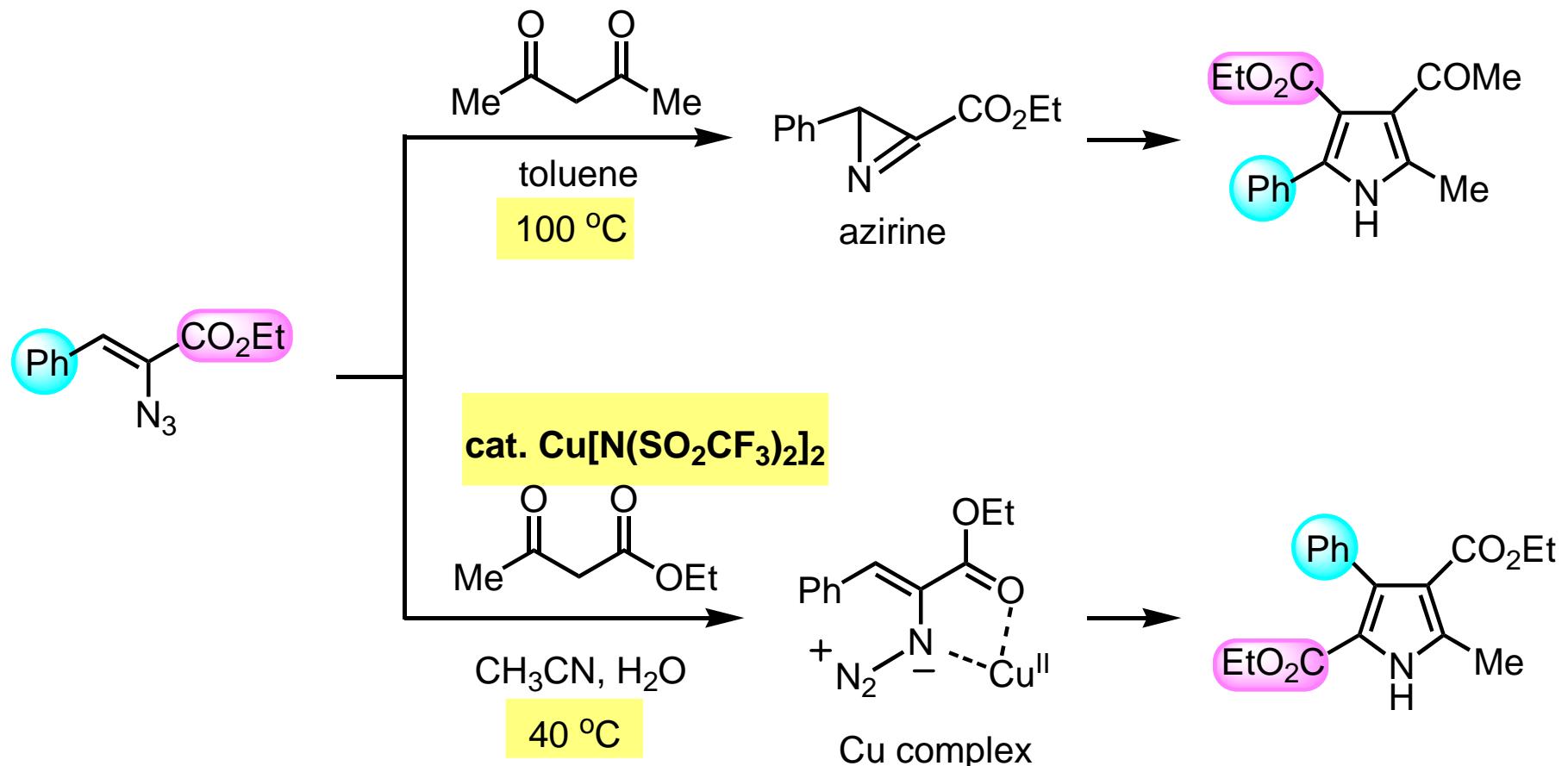
2*H*-Azirine  $\longrightarrow$  Vinyl Azide



D. Knittel, *Synthesis*, **1985**, 186.



# Synthesis of Pyrroles from Vinyl Azides



Org. Lett. 2008, 10, 313.



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## Catalytic C-C and C-N Bond Formation

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